



Simulation and Modeling of the CO₂ Refrigeration System

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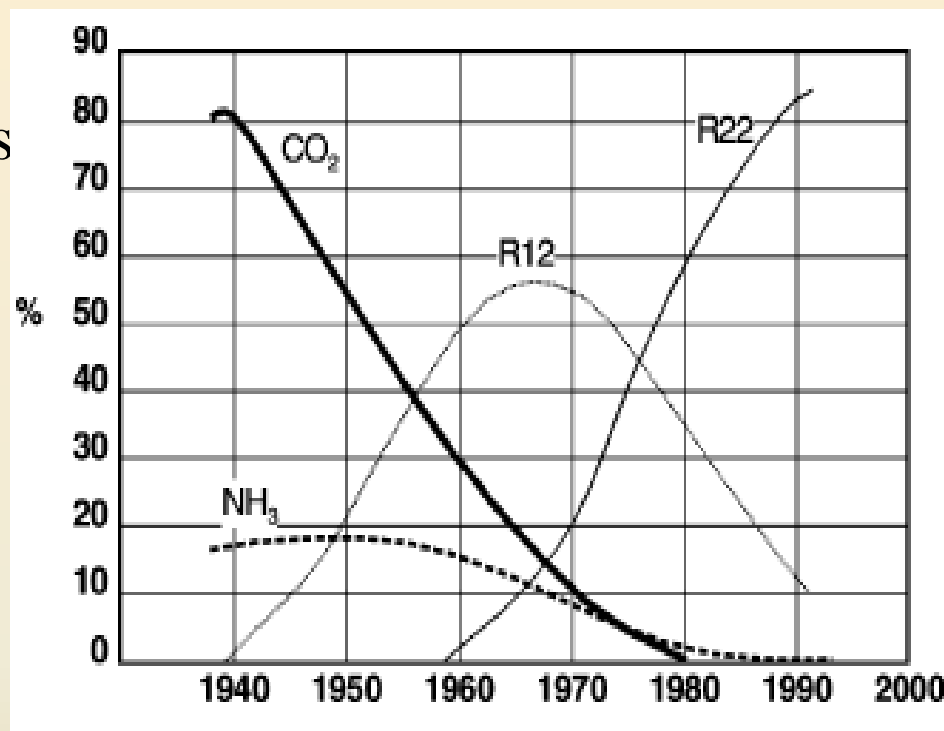
Outline

1. General Background
2. Literature Reviews
3. Simulation Method
4. Simulation results
5. Conclusion



General Background for CO₂ Refrigeration

- Carbon dioxide (CO₂) had been used as a refrigerant for compression refrigeration system as early as 1880.
- Non-toxic, non-combustible, low cost and readily available.
- CO₂ Refrigerant is used over 80% in fishing vessel and relevant transportation.
- Carbon dioxide was greatly phased out when synthetic refrigerant was invented since 1920.
- From 1990, the concerns on environmental concerns has revived the comeback of CO₂.

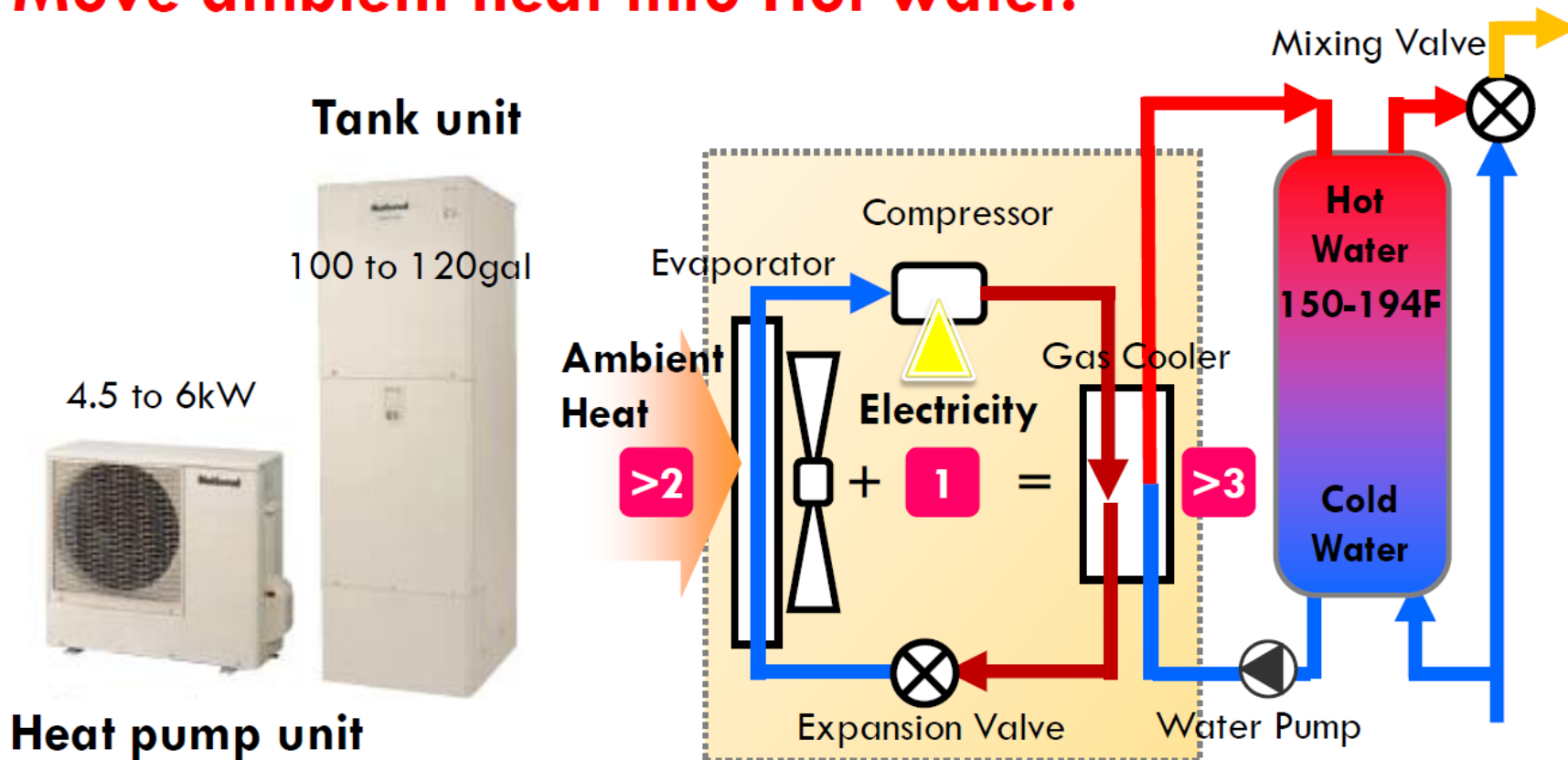


Stera A. Ammonia refrigerating plant on reefer ships. Introduction to ammonia as a marine refrigerant. Lloyd's Register Technical Seminar, London; 1992.



Typical CO₂ Heat Pump Water Heater

Move ambient heat into Hot water.

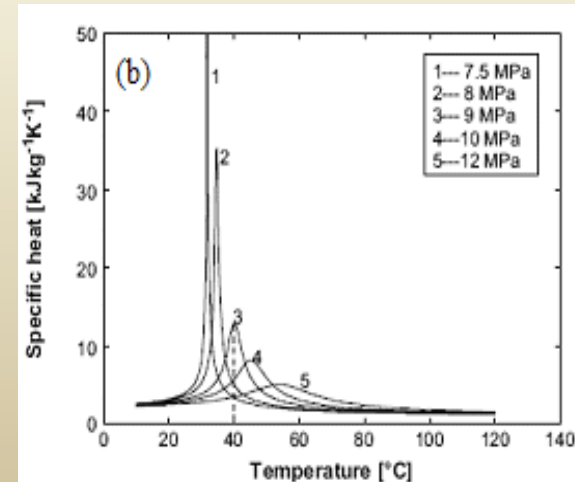
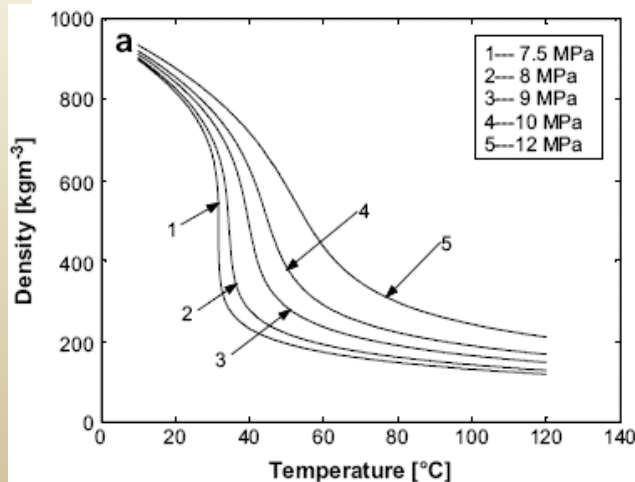
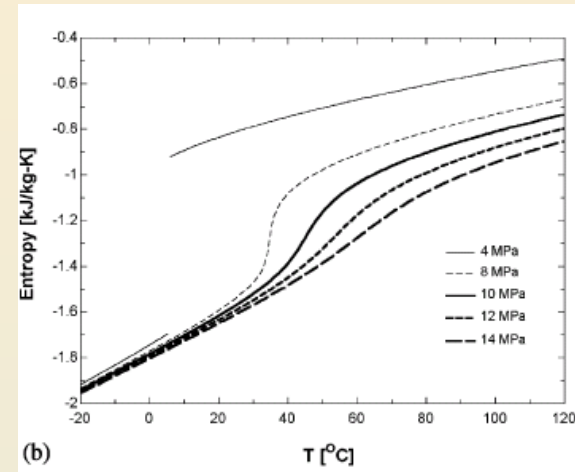
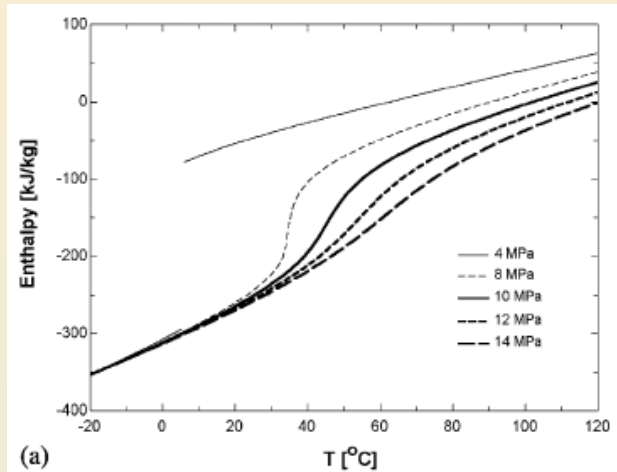


Source of the product picture: model# HE-K37AQS from Panasonic(Matsushita) online catalogue
<http://national.jp/sumai/hp/online.html>



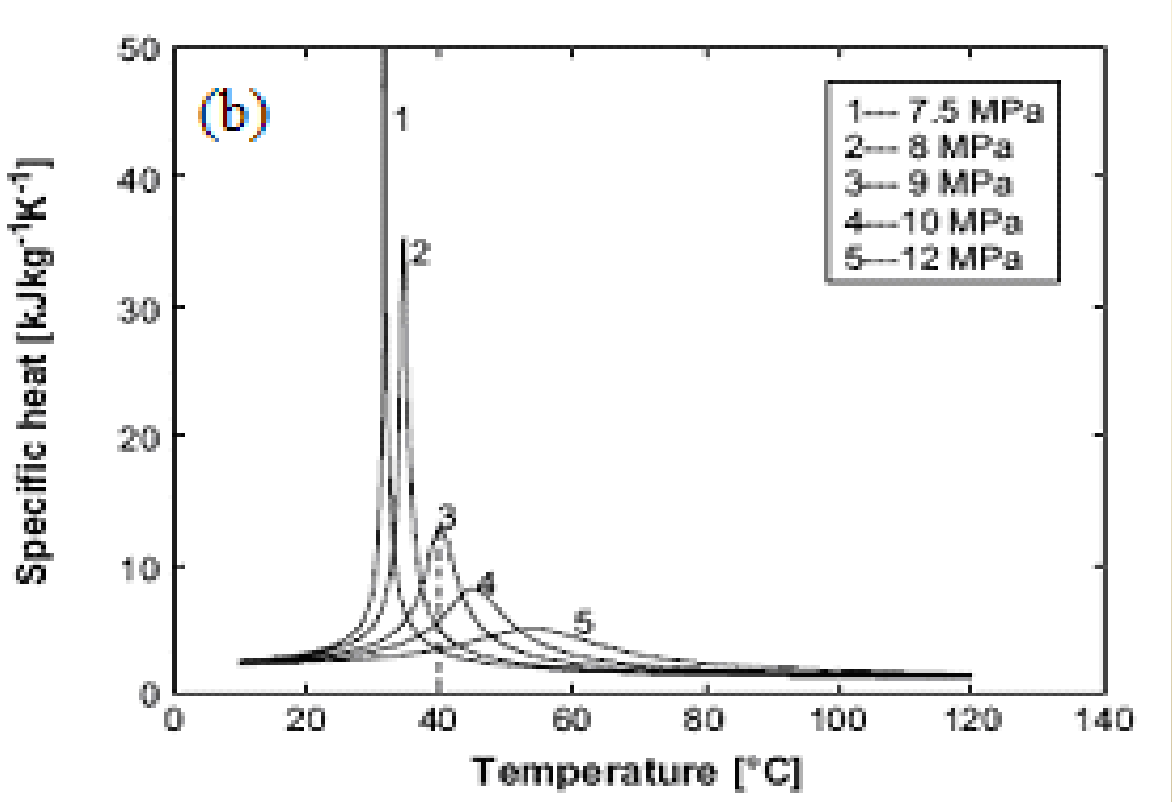
Thermodynamic and transport properties of CO₂ near the critical point

- Near the critical point, density, enthalpy and entropy changes considerably. Specific heat shows a spike phenomenon. The peak is especially pronounced at the “pseudo” critical point.





- When the pressure is increased above the critical pressure, each pressure value corresponds to an extreme value. This temperature with extreme specific heat value is called pseudo-critical temperature. This pseudo-critical temperature will increase as the pressure rises, but the spike of the maximum value of the specific heat becomes less evident with increasing pressure.





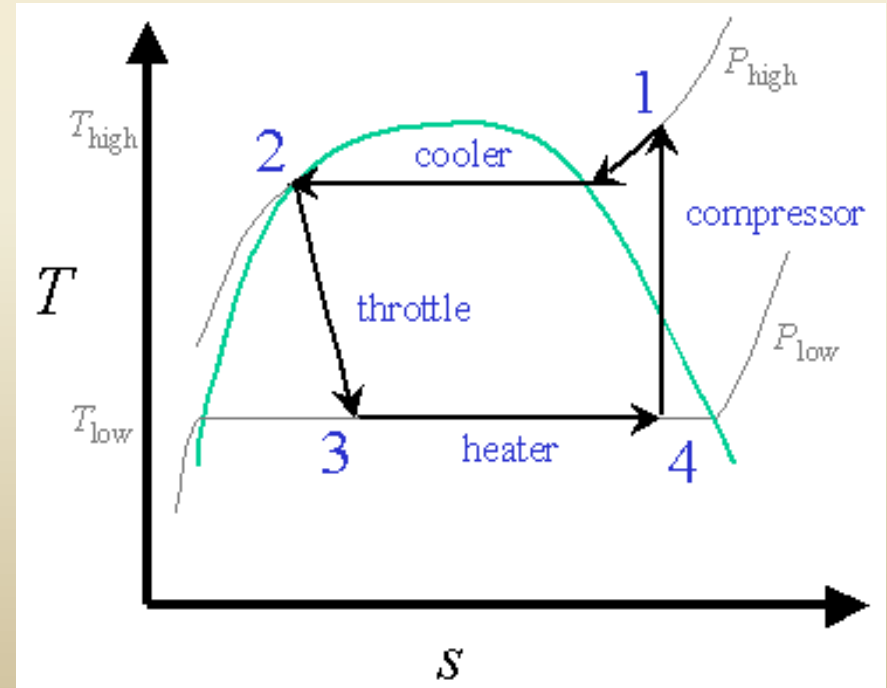
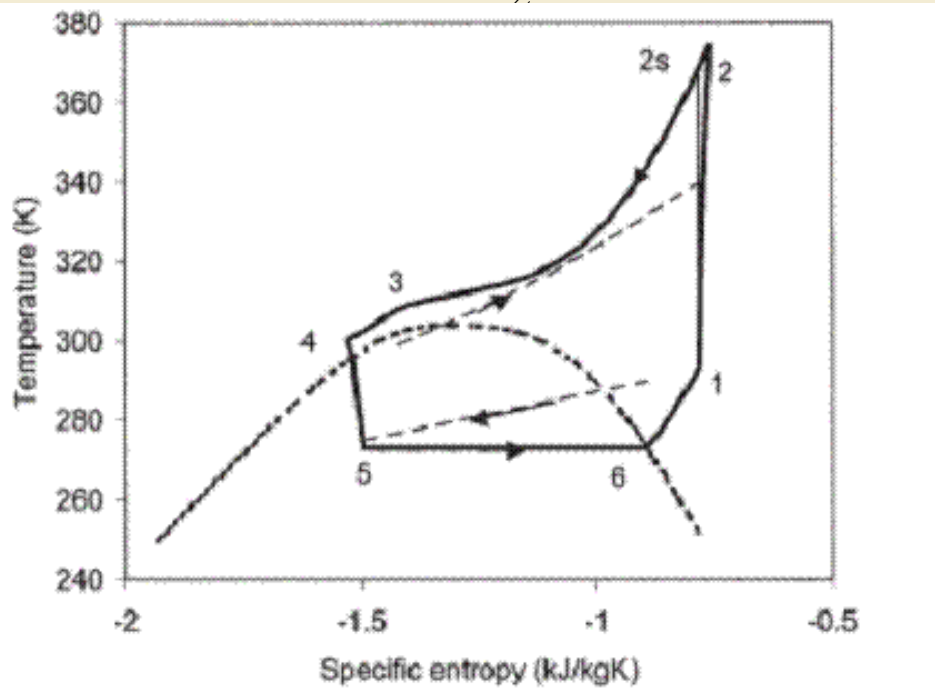
Typical CO₂ refrigerating cycle is above the critical point. The expansion process and the evaporation process are in the subcritical region.

Therefore, CO₂ refrigeration cycle operates across supercritical and subcritical point region and they are called the **Transcritical Cycle**.

Because the temperature of the critical point of carbon dioxide is low (about 31.1 °C). Therefore, for the normal ambient temperature, no condensing occurs and single-phase gas cooling is encountered.

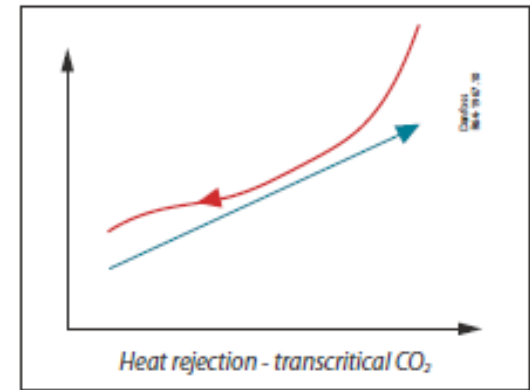
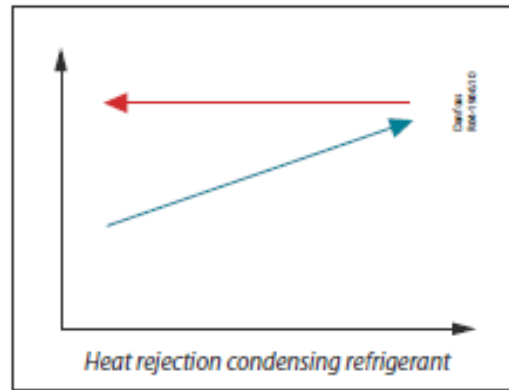
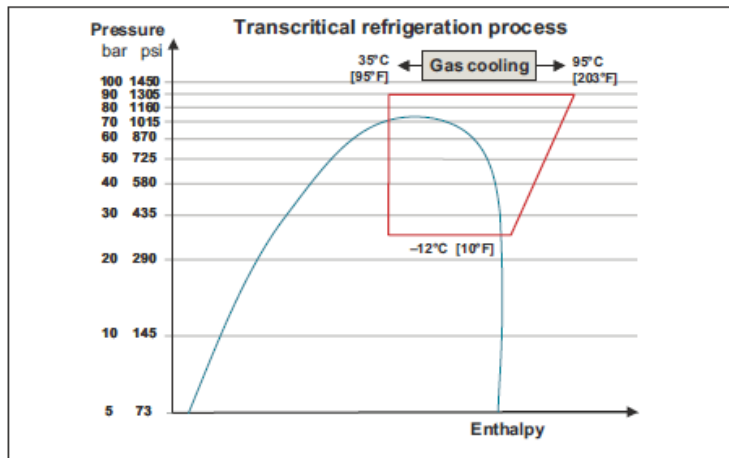
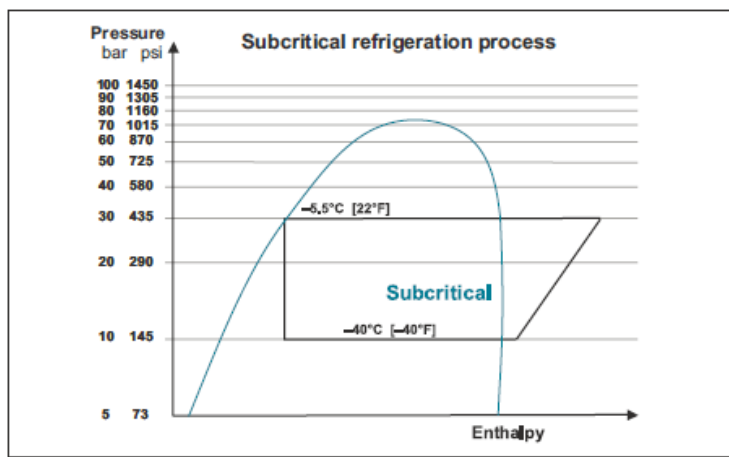
CO₂

R-134a





Typical P-h diagram for CO₂ & conventional refrigerant





COP of the Refrigerants

Table 1. Characteristics of some synthetic and natural refrigerants

Applied Thermal Engineering Vol. 17, No. 1, pp. 33–42, 1997

| Refrigerant | R-12 (CFC) | R-22 (HCFC) | R-134a (HFC) | R-717 (NH ₃) | R-744 (CO ₂) | R-290 (propane) | R-600 (butane) | R-718 (H ₂ O) | R-728 (air) |
|--|------------|-------------|--------------|--------------------------|--------------------------|-----------------|----------------|--------------------------|-------------|
| Natural substance | no | no | no | yes | yes | yes | yes | yes | yes |
| ODP ^a | 0.9 | 0.05 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| GWP ^b | 3 | 0.34 | 0.29 | 0 | 0 ^c | < 0.03 | < 0.03 | 0 | 0 |
| Toxicity TLV (ppm, volume) ^d | 1000 | 500 | 1000 | 25 | 5000 | 1000 | 1000 | no | no |
| Flammability | no | no | no | yes | no | yes | yes | no | no |
| Critical point temperature (°C) | 115.5 | 96.2 | 100.6 | 133 | 31.1 | 96.8 | 152.1 | 374.2 | -140 |
| Critical point pressure (bar) | 40.1 | 49.9 | 40.7 | 114.2 | 73.7 | 42.6 | 38.0 | 221.2 | 37.2 |
| Normal boiling point (°C) | -30 | -40.8 | -26 | -33.3 | -78.4 | -42.1 | -0.4 | 100 | no |
| Maximum refrigeration capacity at 0°C (kJ/m ³) | 2733 | 4344 | 2864 | 4360 | 22 600 | 3888 | 1040 | 1349 ^e | — |

^aOzone depletion potential—compared with R-11.

^bGlobal warming potential—compared with R-11.

^cThreshold limit value for exposure of 8 h/day, 40 h/week, without any adverse effect.

^dZero effective GWP, because more than sufficient quantities of it can be recovered from waste gases.

^eAt 100°C.

Table 1 – Comparative refrigerant performance

| No. | Name | CoP |
|--------|-------------------------------|------|
| R-717 | Ammonia | 4.84 |
| R-290 | Propane | 4.74 |
| R-600 | Butane | 4.68 |
| R-22 | Chlorodifluoromethane | 4.65 |
| R-134a | Tetrafluoroethane | 4.60 |
| R-407C | R-32/R-125/R-134a (23/25/52) | 4.51 |
| R-410A | R-32/R-125 (50/50) | 4.41 |
| R-404A | R-125/R-143a/R-134a (44/52/4) | 4.21 |
| R-744 | Carbon dioxide | 2.96 |

Based upon a standard operating cycle of 258 K evaporating temperature, 303 K condensing temperature, 0 K subcooling and 0 K superheat.

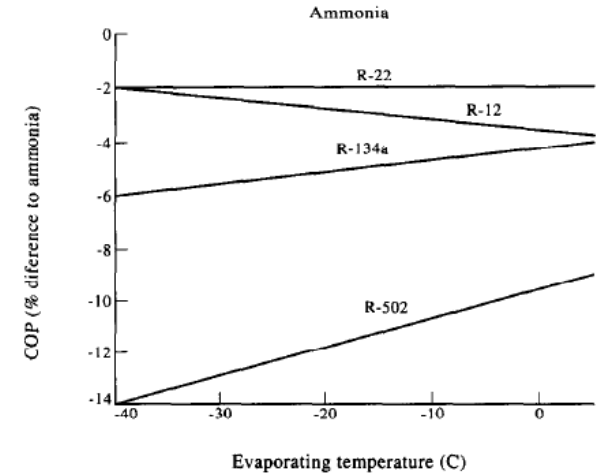


Fig. 2. Relative COP of an isentropic vapour compression cycle for different refrigerants [4].

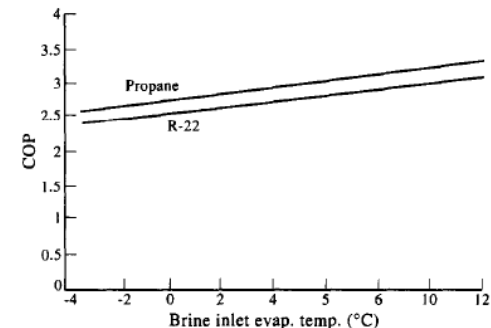


Fig. 3. COP of a heat pump with R-22 and R-290, for different brine inlet evaporator temperatures [12].



Comparison between CO₂, R410A & R-407C

Table 1. Refrigerant characteristics

| Refrigerant | HFC | | Natural refrigerant |
|---|---------------------------------------|---|---|
| | R410A | R407C | R744(CO ₂) |
| Practical examples of commercialization | RAC PAC Commercial Water Heater | PAC Chiller Commercial Water Heater | Residential Water Heater (Eco Cute) & Commercial Water Heater |
| ODP ¹ | 0 | 0 | 0 |
| GWP ² | 1975 | 1652.5 | 1 |
| Combustible | No | No | No |
| Toxic | Low | Low | Low |
| Pressure (MPa) (low/ high) | 2.7/3.0 | 1.8/2.0 | 9.5/11 |
| COP(compared to R410A) (low/high) | 100 | 95/100 | 60/80 |

Note: RAC= residential air-conditioning; PAC= packaged air-conditioning.

¹ Ozone depletion potential

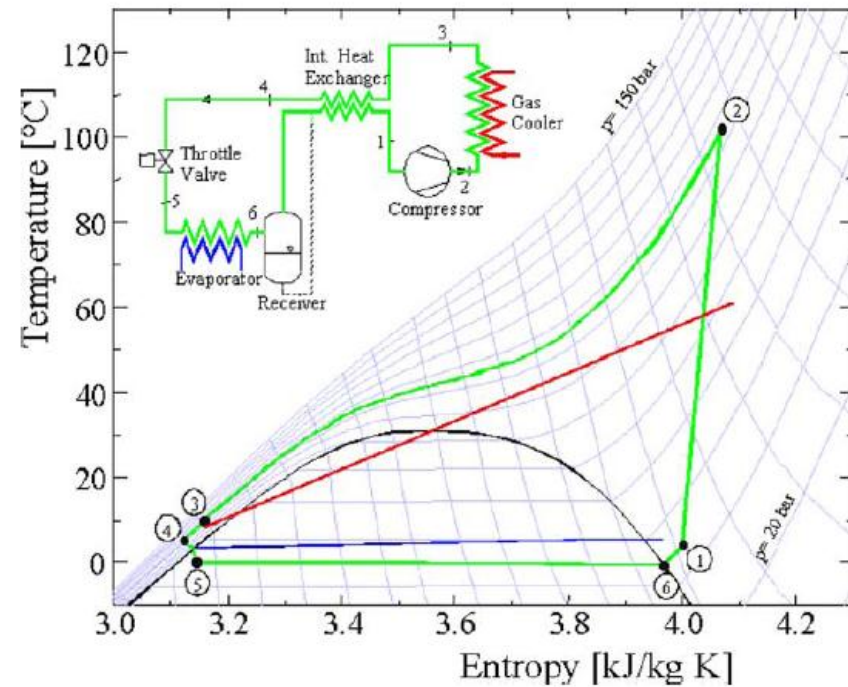
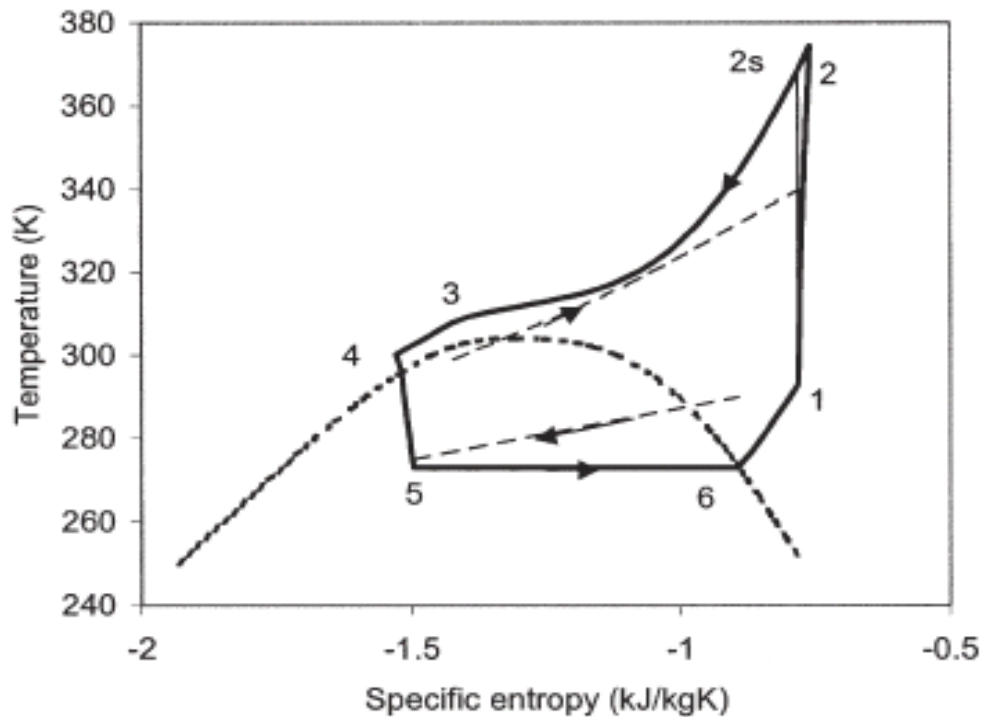
² Global warming potentials are based on IPCC 2001.

2010 International Symposium on Next-generation Air Conditioning and Refrigeration Technology,
17 – 19 February 2010, Tokyo, Japan

The Development of Heat Pump Water Heaters Using CO₂ Refrigerant



Aim of this study: develop a simulation program for CO₂ refrigeration cycle.





Literature Reviews

- There is a big difference between carbon dioxide refrigeration cycle and traditional refrigerants. The critical temperature and critical pressure of CO₂ are **31.1°C** and **78.8 Bar** in room temperature applications (such as outside air at 35 °C), and the refrigerant will operate above the critical point.
- CO₂ refrigerant can't be cooled by ordinary condenser. CO₂ is cooled by near single-phase gas. Normally we call the heat exchanger as **Gas Cooler** rather than condenser.

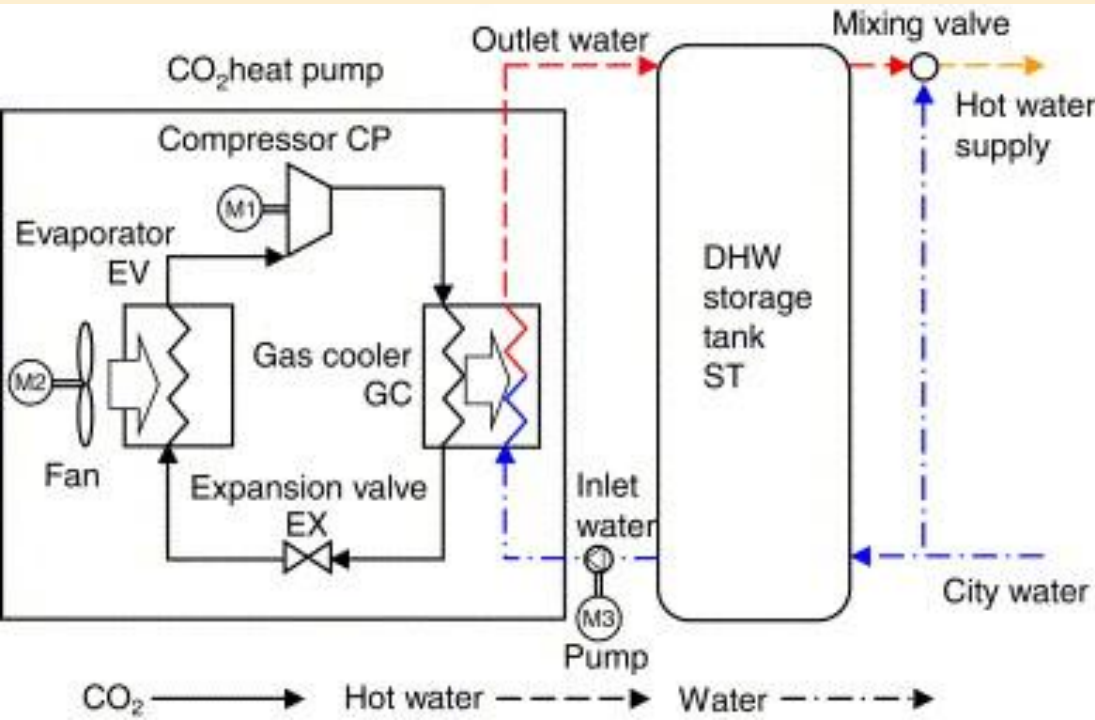


| | Gas Cooler | Evaporator | Expansion device Model | Inner HX | Method (conditions outside the HX geometry, the water inlet temperature, air inlet temperature) |
|------------------|--------------------------|--------------------------|---|----------|---|
| Kim et al. | Water Tube-in-tube | Water Tube-in-tube | $\Delta h=0$ | yes | Fixed: Pdis、Tsup |
| Yang et al. | Water Shell-tube | Water Shell-tube | $\dot{W}_{exp} = G_r (h_3 - h_{4,is}) \times \eta_{exp,is} \times \eta_{exp,m}$ | No | Fixed: Pdis |
| Sarker et al. | Water Tube-in-tube | Water Tube-in-tube | $\Delta h=0$ | Yes | Maximum COP |
| Yokoyama et al. | Water Tube (double-pipe) | Water Tube (double-pipe) | $\Delta h=0$ | No | Fixed: \dot{m}_{co2} 、 P_{eva} 、 $T_{ci,gc}$ |
| Yamaguchi et al. | Water Tube-in-tube | Air Fin-tube | $\Delta h=0$ | yes | Fixed : $T_{w,o}$ |
| Wang et al. | Water Tube-in-tube | Water Tube-in-tube | $\Delta h=0$ | no | Maximum COP T_{sup} 、 $T_{w,o}$ |
| Sarker et al. | No description | No description | $\Delta h=0$ | yes | Fixed Maximum COP |
| Sarker et al. | Water Tube-in-tube | Water Tube-in-tube | $\Delta h=0$ | No | Fixed :Pdis Fixed :degree of superheat |
| Sarker et al. | Water Tube-in-tube | Water Tube-in-tube | $\Delta h=0$ | No | Fixed: Pdis Fixed :Psuc |

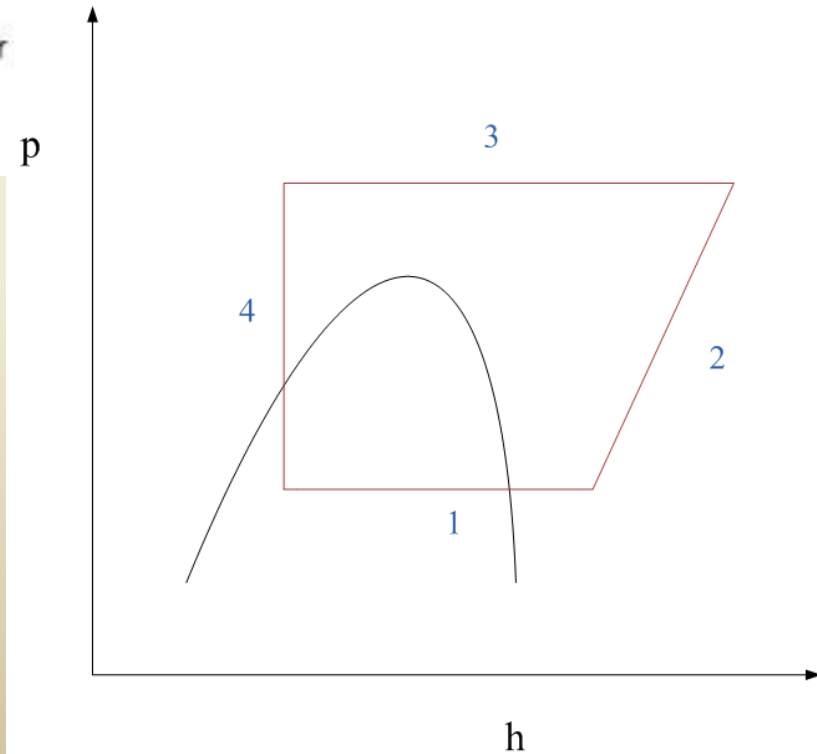


Basic outline of the present simulation method

| | Gas Cooler | Evaporator | Expansion device Model | Inner HX | Method (conditions outside the HX geometry, the water inlet temperature, air inlet temperature) |
|---------------|--------------------|-----------------------------|---|----------|---|
| Present Study | Water Tube-in-tube | Air Fin-and-Tube (Wet Coil) | $\Delta h = 0$ Capillary pressure drop equation | No | No Fixed conditions |



The CO₂ refrigeration Cycle contains four major components, including a gas cooler, an evaporator, a compressor and an expansion valve.





Compressor Model

The compressor calculations assume for the isentropic adiabatic process. First, we can generate equation by the continuity equation and the energy equation.

$$\dot{m}_{r,o} - \dot{m}_{r,i} = 0$$

$$\dot{m}_{r,o} h_{r,o} - \dot{m}_{r,i} h_{r,i} = W_{com}$$

isentropic efficiency and volumetric efficiency for the compressor

$$\eta_{isen} = \frac{h_{r,isen} - h_{r,i}}{h_{r,o} - h_{r,i}} \quad \eta_v = \frac{\dot{m}_{r,i}}{\rho_{r,i} V_{com} N_{com}}$$

From Sarkar et al.,

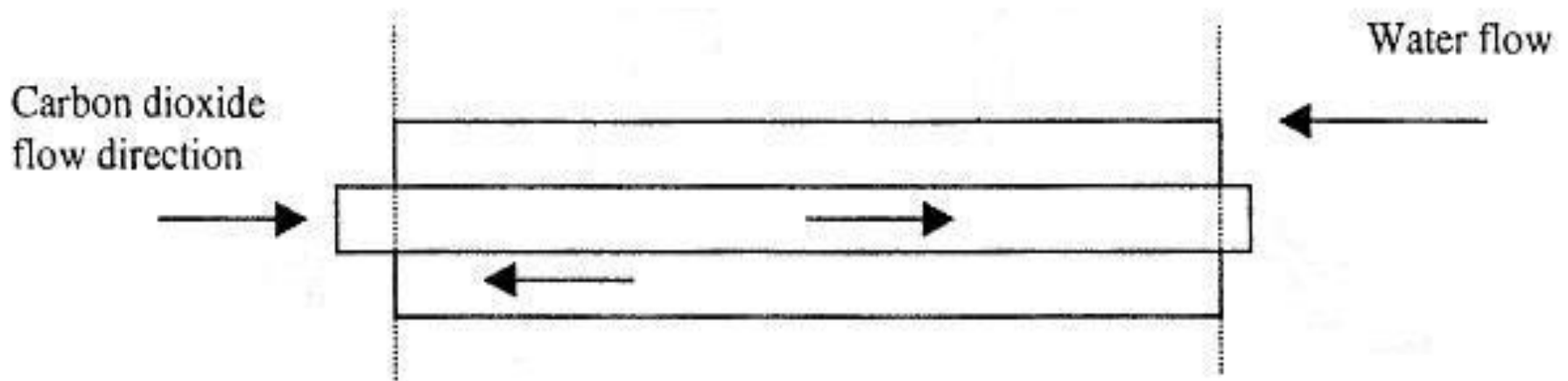
$$\eta_v = 0.9207 - 0.0756 \left(\frac{P_{dis}}{P_{suc}} \right) + 0.0018 \left(\frac{P_{dis}}{P_{suc}} \right)^2$$

$$\eta_{isen} = -0.26 + 0.7952 \left(\frac{P_{dis}}{P_{suc}} \right) - 0.2803 \left(\frac{P_{dis}}{P_{suc}} \right)^2 + 0.0414 \left(\frac{P_{dis}}{P_{suc}} \right)^3 - 0.0022 \left(\frac{P_{dis}}{P_{suc}} \right)^4$$



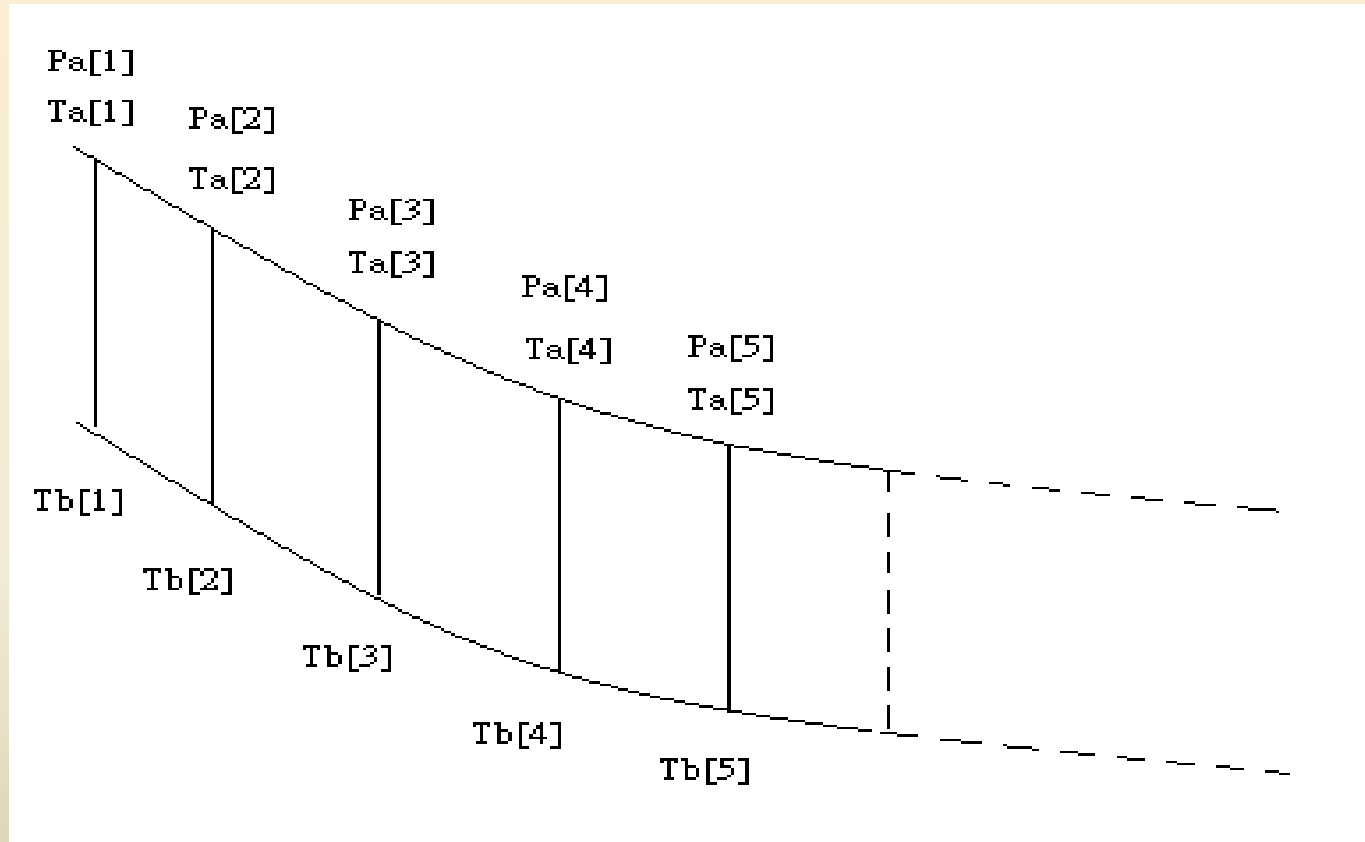
Objective: Simulation of the Targeted HX – A Tube-in-tube HX (Double Pipe HX)

- The heat exchanger is a tube-in-tube type, and the water flows counter-currently against the coolant (carbon dioxide) during the heat exchanging process.





Discretizing the HX due to significant change of physical properties



Due to considerable change of physical properties, especially near pseudo-critical temperature, the heat exchanger must be subdivided into many small segments.



Basic equations

- Energy balance/pressure drop amid water and coolant can be written in the following equations:

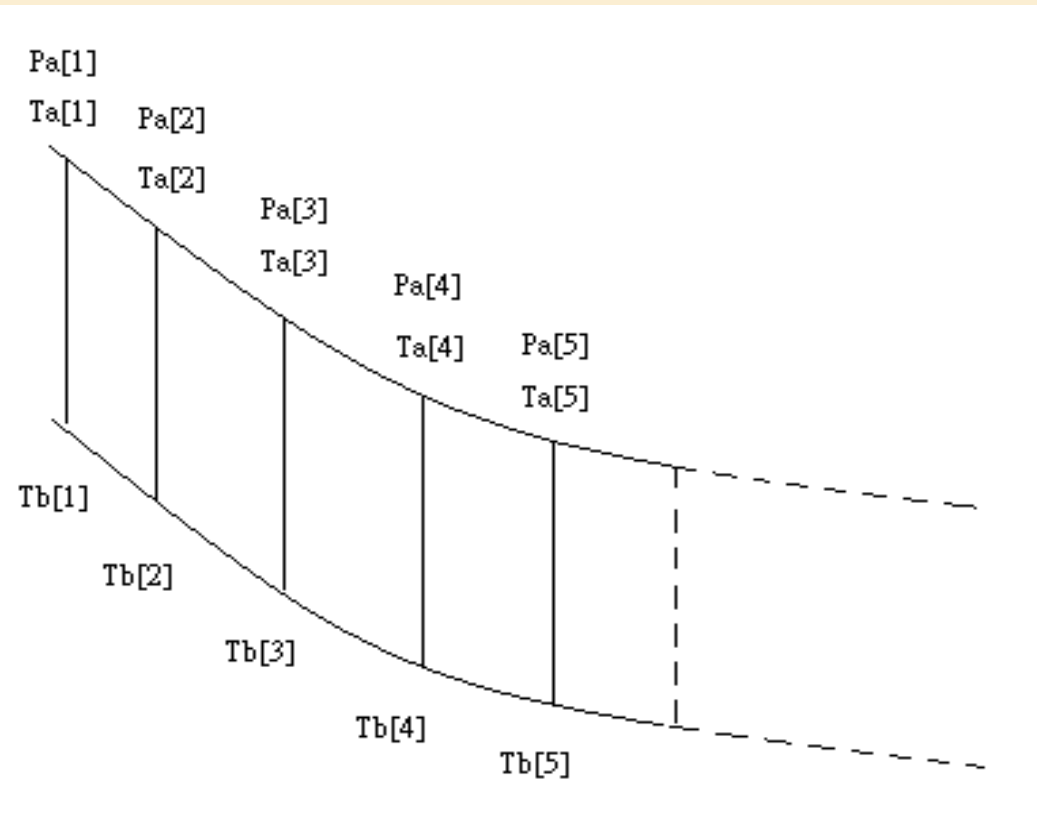
$$Q_1 = m_c \times Cp_{c1} \times (T_{a1} - T_{a2}) = m_w \times Cp_{w1} \times (T_{b1} - T_{b2}) \quad (1)$$

$$Q_1 = (UA)_1 \times (LMTD)_1 \quad (2)$$

$$P_{a2} - P_{a1} = \frac{4L}{d} \times f \times \frac{G_c^2}{2\rho_c} \quad (3)$$



Physical Configurations



Each segment has three unknown parameters. (water temperature, CO₂ temperature and CO₂ pressure).

There are three equations for each segment.

The unknown parameters can thus be solved through these equations.



HTCs in CO₂ & water

- For CO₂, Dang and Hihara (2004) correlation :

$$N_{u} = \frac{\left(\frac{f}{s}\right)(Re-1000)Pr}{1.07 + 12.7 \sqrt{\frac{f}{s}}(Pr^{2/3}-1)} \quad Pr = \begin{cases} Cp_b \mu_b / \lambda_b, & \text{for } Cp_b \geq \overline{Cp} \\ \overline{Cp_b} \mu_b / \lambda_f, & \text{for } Cp_b < \overline{Cp} \text{ and } \mu_b / \lambda_b \geq \mu_f / \lambda_f \\ \overline{Cp_b} \mu_f / \lambda_f, & \text{for } Cp_b < \overline{Cp} \text{ and } \mu_b / \lambda_b < \mu_f / \lambda_f \end{cases}$$

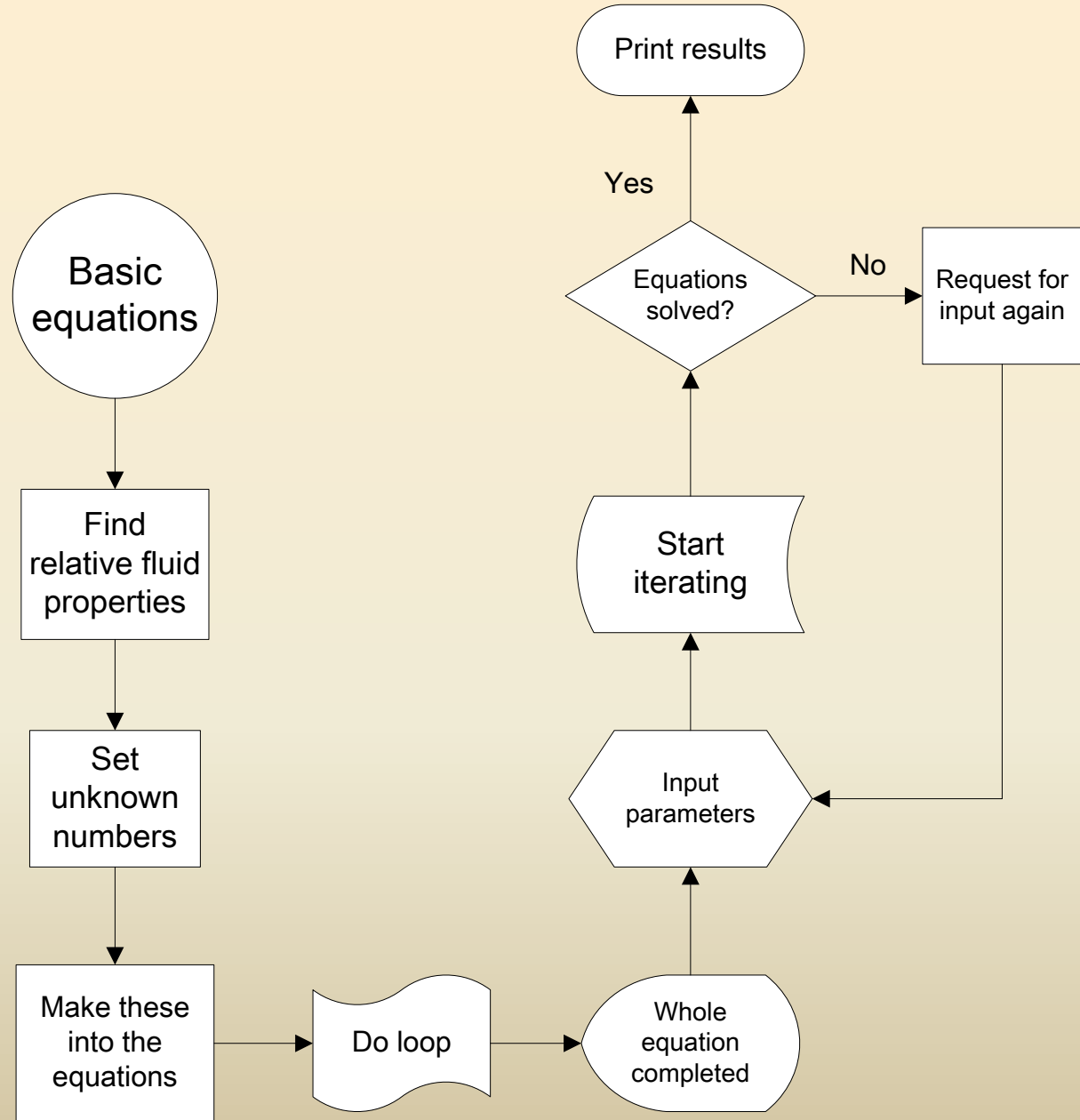
- For water, Genielinski 方程式 :

$$N_{u} = \frac{\left(\frac{f}{2}\right)(Re-1000)Pr}{1.07 + 12.7 \sqrt{\frac{f}{2}}(Pr^{2/3}-1)} \quad f = [1.58 \ln(Re_b) - 3.28]^{-2}$$



Algorithm for Solution of the Gas Cooler

1. List the basic equations.
2. Make the properties of fluid into these equations.
3. Decide the segment number for calculation, and build the completed equation sets through do loops.
4. Start iterating.
5. Print out the results if the equations are convergent.



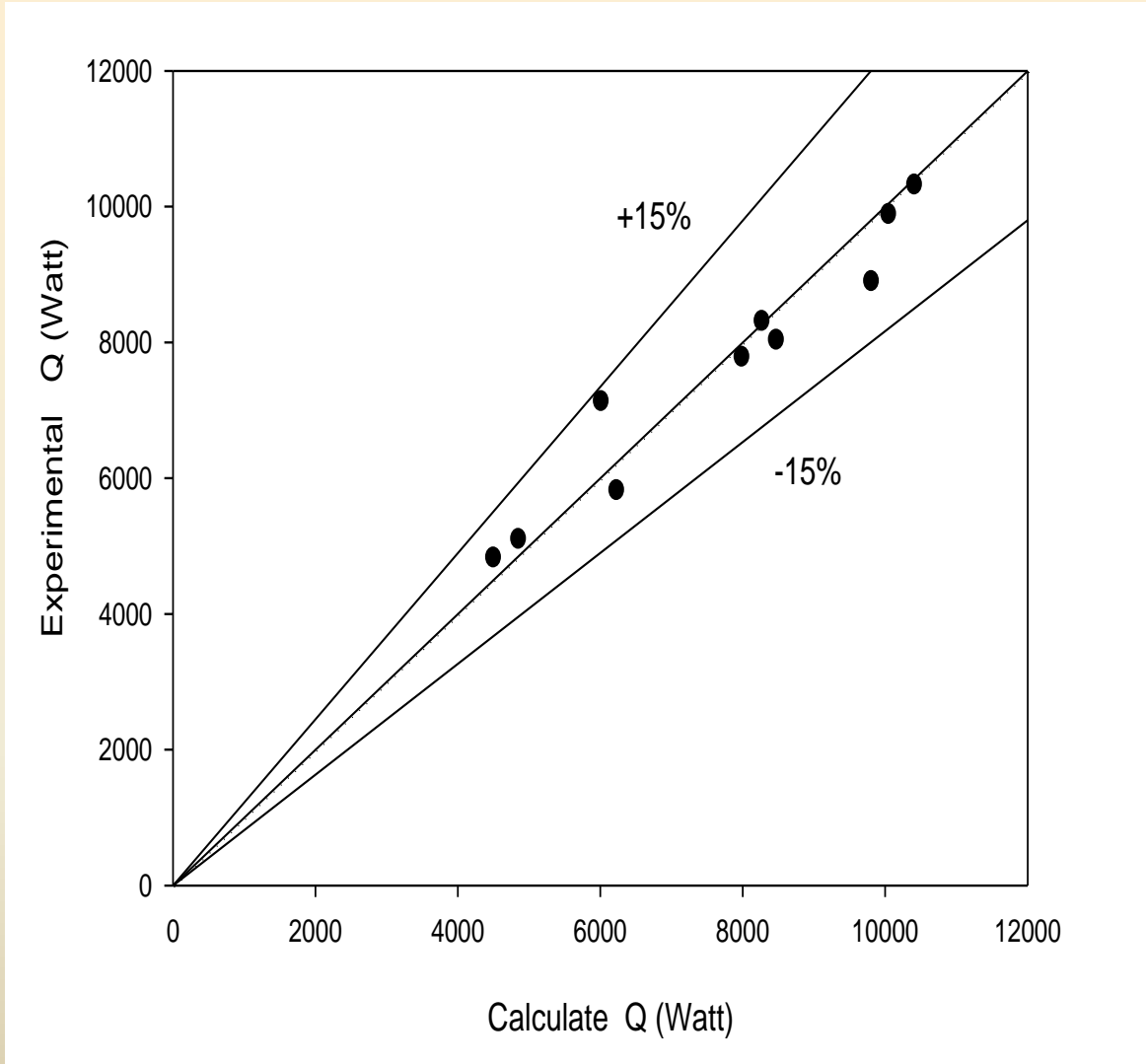
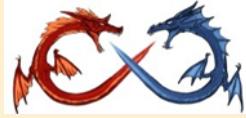


Results and Discussion

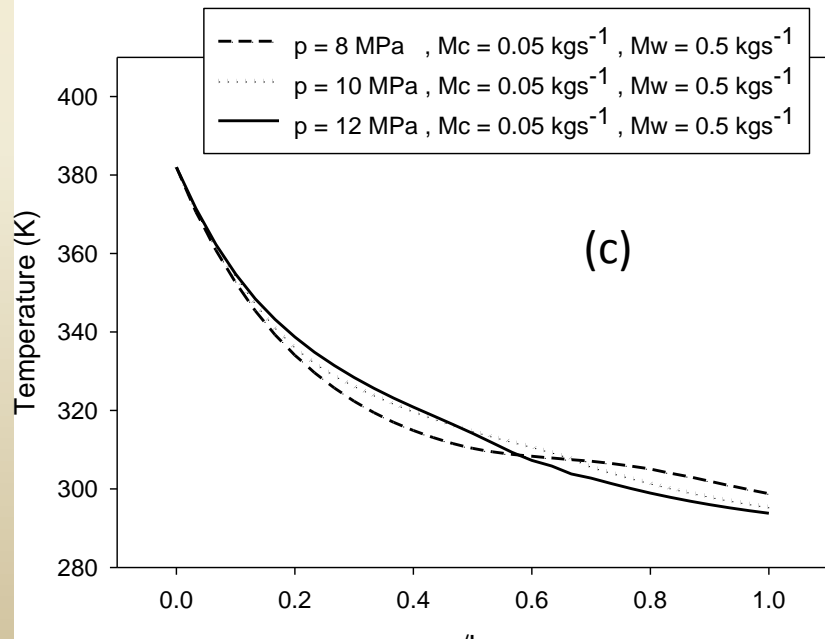
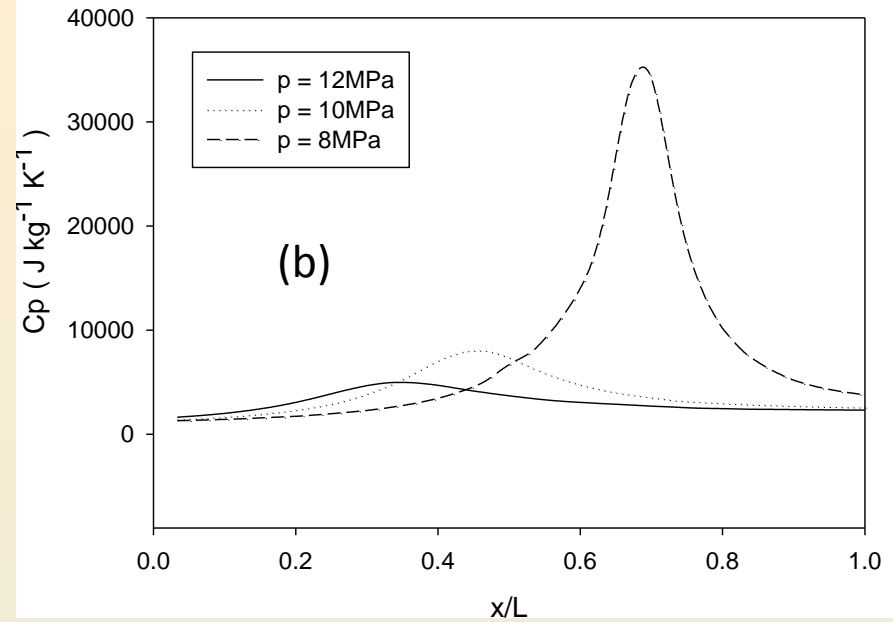
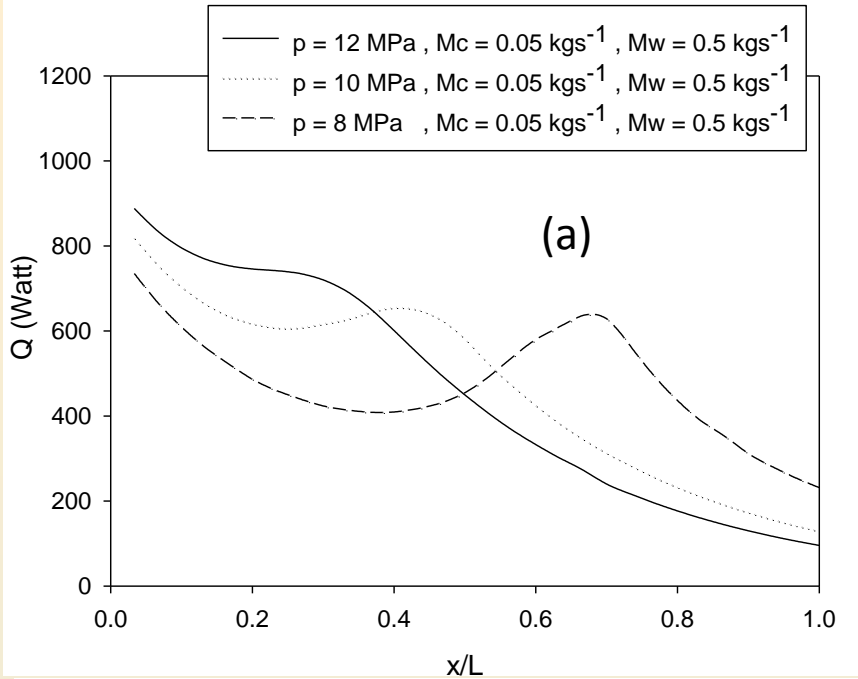
- For validating the proposed model, calculation is compared with the measurements of Pitla et al.

| | $T_{c,in}$ (°C) | $P_{c,in}$ (Mpa) | $T_{w,in}$ (°C) | Mc (kg/s) | Mw (kg/s) |
|-------|-----------------|------------------|-----------------|-------------|-------------|
| Run1 | 121.2 | 9.44 | 20.8 | 0.01963 | 0.04011 |
| Run2 | 126 | 11.19 | 24.2 | 0.0274 | 0.040497 |
| Run3 | 73.3 | 13.33 | 36.12 | 0.02043 | 0.12914 |
| Run4 | 123.5 | 10.8 | 27.21 | 0.02862 | 0.084087 |
| Run5 | 107.2 | 8.11 | 24.2 | 0.0198 | 0.0455 |
| Run6 | 123.4 | 8.98 | 22.3 | 0.02996 | 0.067864 |
| Run7 | 118.3 | 7.79 | 21.2 | 0.02123 | 0.066434 |
| Run8 | 115.8 | 8.60 | 18.9 | 0.03436 | 0.084087 |
| Run9 | 114.9 | 8.76 | 18.9 | 0.03638 | 0.065091 |
| Run10 | 113.4 | 9.50 | 15.9 | 0.03825 | 0.109052 |

Pitla et al. had conducted CO₂ tube-in-tube heat exchanger with ID = 0.00472m and OD = 0.00635m for inner tube, ID = 0.01575m for outer tube.



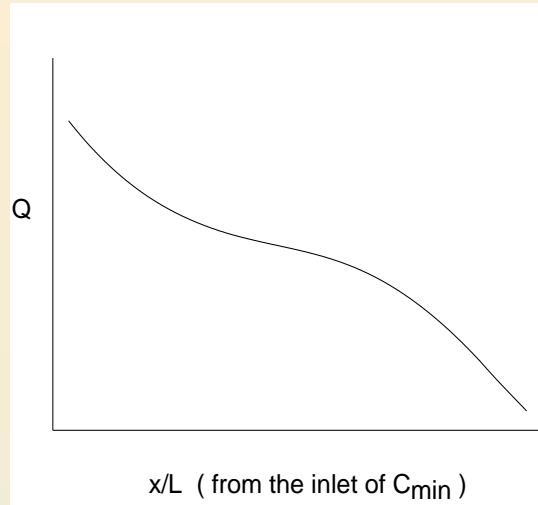
The calculations are in line with the experimental measurement, suggesting the applicability of the present modeling.



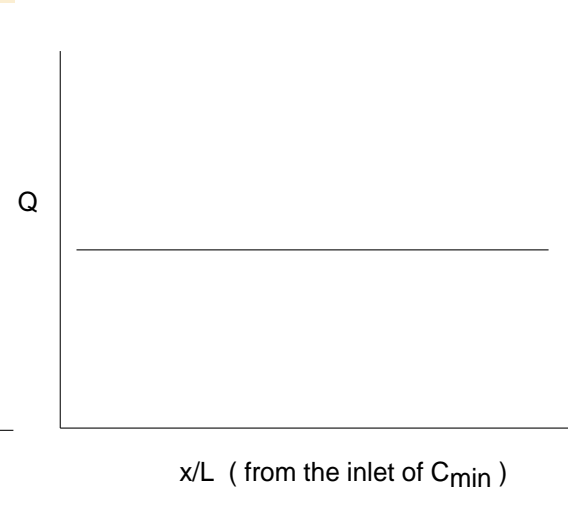
Effect of the inlet pressure on (a) the variation of the local heat transfer rate vs. the dimensionless tube length, (b) the variation of c_p vs. the dimensionless tube length and (c) the variation of the CO_2 temperature vs. the dimensionless tube length.



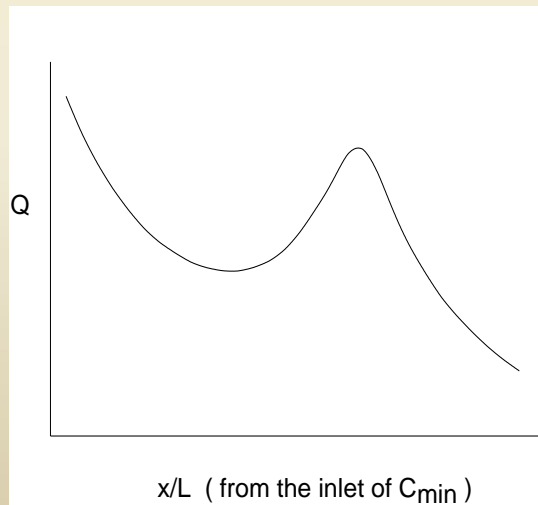
Schematic of the variation of the local heat transfer rate for a tube-in-tube heat exchanger: (a) Constant property, $C_{\min} \neq C_{\max}$; (b) Constant property, $C_{\min} = C_{\max}$; (c) CO_2 flow across the pseudo-critical point; and (d) Variable property.



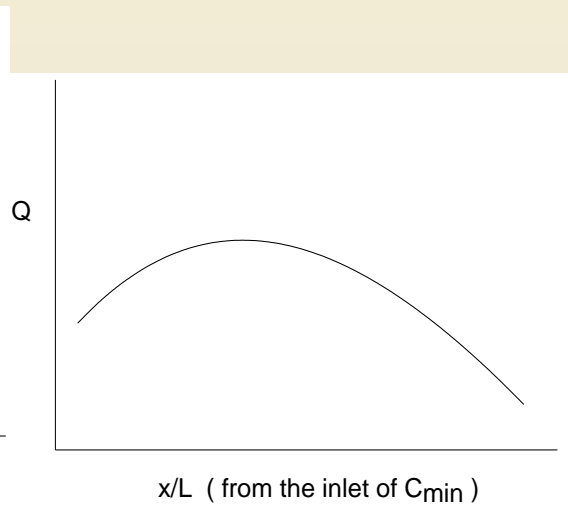
(a)



(b)



(c)



(d)



Concluding remarks for Gas Cooler

- Unlike conventional working fluid which normally shows a monotonically decrease of local heat transfer rate along the tube length, the CO₂ shows a different trend as compared to typical sub-critical fluids.
- The local heat transfer rate does not monotonically decrease with the tube length.
- In fact, a plateau occurs somewhere inside the heat exchanger. Moreover, a second maximum is seen when p is below 10 MPa, yet a significant of recovering of local heat transfer rate is encountered for $p = 8$ MPa despite the maximum temperature difference still occurs at the water inlet.
- The phenomenon is attributed to the significant rise of specific heat of CO₂ near pseudo critical temperature.
- The effect of tube diameter plays a crucial role of CO₂ HX, this is especially pronounced when the thermal resistance is on the CO₂ side.



Capillary tube Model



The calculation condition is based on the continuity equation, energy equation and pressure drop equations:

$$\dot{m}_{r,o} - \dot{m}_{r,i} = 0$$

$$\dot{m}_{r,o} h_{r,o} - \dot{m}_{r,i} h_{r,i} = 0$$

Agrawal et al. [14] pressure drop equation :

$$\frac{dp}{dL} = \left(\frac{dp}{dL} F\right) + \left(\frac{dp}{dL} a\right) \quad \frac{dp}{dL} = -G^2 \left(f_{tp} \frac{v}{2D} + \frac{dv}{dL}\right)$$

Lin et al. [15] empirical formula can calculate the friction coefficient of the two-phase flow :

$$f_{tp} = \phi_{tp} f_{sp} \left(\frac{v_{sp}}{v_{tp}}\right) \quad \phi_{tp} = \left[\frac{(8/\text{Re}_{tp})^{12} + (A_{tp}^{16} + B_{tp}^{16})^{-3/2}}{(8/\text{Re}_{sp})^{12} + (A_{sp}^{16} + B_{sp}^{16})^{-3/2}} \right]^{1/12} \left[1 + x \left(\frac{v_g}{v_l} - 1\right) \right]$$

$$A = 2.457 \ln \frac{1}{(7/\text{Re}_{tp})^{0.9} + 0.27 \varepsilon / D} \quad B = \frac{37530}{\text{Re}_{tp}}$$



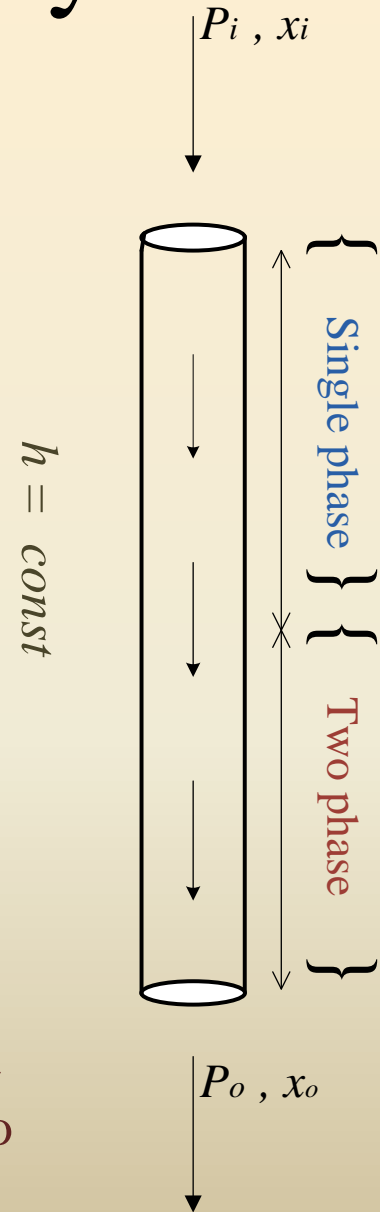
pressure drop in capillary tube

The refrigerant flow out from the condensation region and through a smooth tube in single-phase liquid. It's change to two-phase in the tube in constant enthalpy.

$$\frac{1}{\mu_{tp}} = \frac{1-x}{\mu_l} + \frac{x}{\mu_g}$$

$$\frac{dp}{dL} = -G^2 \left(f_{tp} \frac{v}{2D} + \frac{dv}{dL} \right)$$

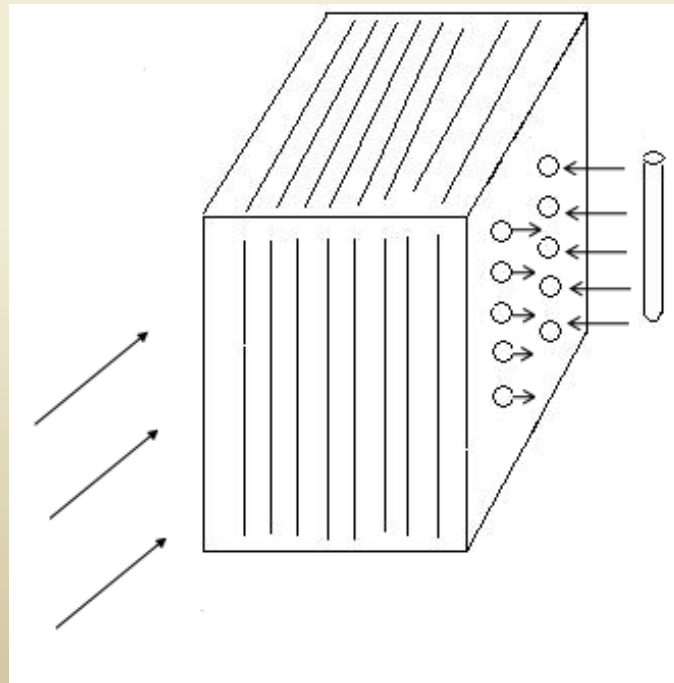
The capillary tube is further divided into many small segments to calculate pressure (Δp) and quality(x) alongside the capillary tube. The outlet quality and exit pressure is also obtained.





Evaporator Model

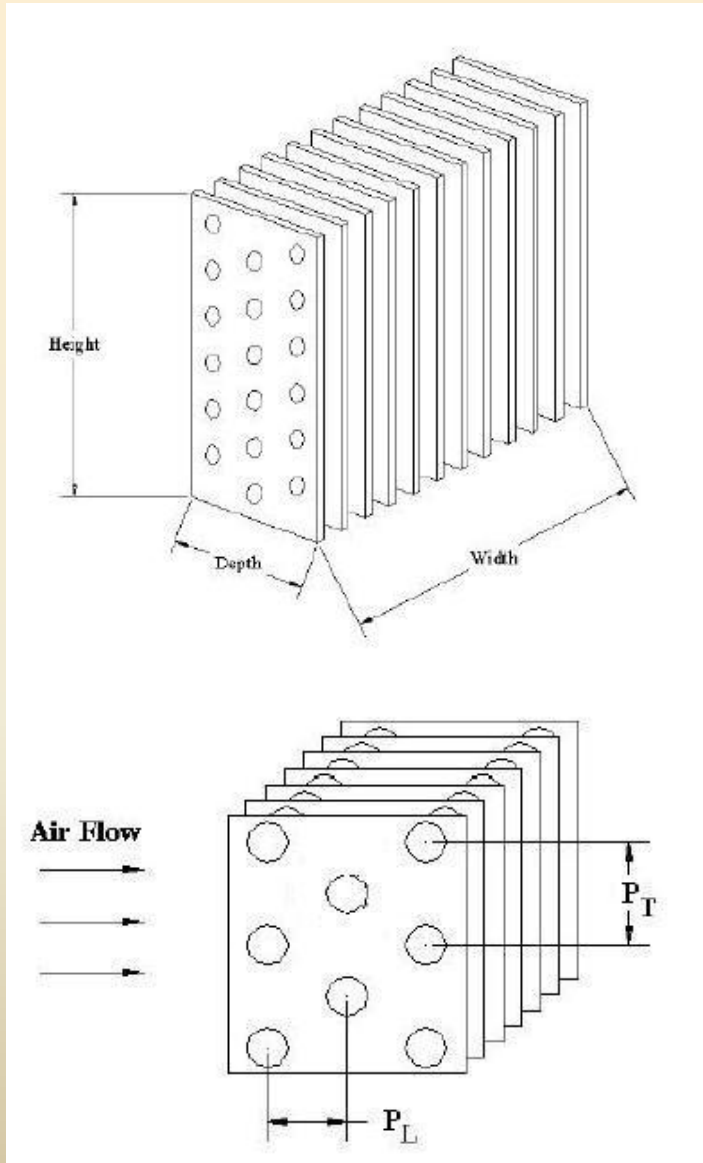
- Fin-tube heat exchanger in evaporator region. Fluid flow pattern is a counter-cross flow. CO₂ refrigerant is flowing in the tube and the air is flowing outside the tube.



Wet coil Analysis
Method



Geometric parameters



| | |
|-------------------------|------------|
| Heat exchanger width | W |
| Heat exchanger height | H |
| Tube diameter | d_o |
| Fin type | Flat type |
| Pitch | F_p |
| Horizontal pitch | P_l |
| Vertical pitch | P_t |
| Fin thickness | δ_f |
| Wall material | Copper |
| Fin material | Aluminum |
| Tube row number | N |
| each row number of tube | N_t |



Threlkeld (1970) for wet coil analysis:

$$Q_a = \dot{m}_a (i_{a,i} - i_{a,o}) \quad (\text{Air side})$$

$$Q_c = \dot{m}_c C_{p,c} (T_{c,o} - T_{c,i}) \quad (\text{Refrigerant side})$$

$$Q = U_{o,w} A_o F \Delta i_m$$

$$\Delta i_m = i_{a,i} + \frac{(i_{a,i} - i_{a,o})}{\ln \left(\frac{i_{a,i} - i_{r,o}}{i_{a,o} - i_{r,i}} \right)} - \frac{(i_{a,i} - i_{a,o})(i_{a,i} - i_{r,o})}{(i_{a,i} - i_{r,o}) - (i_{a,o} - i_{r,i})}$$

$i_{a,i}, i_{a,o}$: inlet and outlet of air enthalpy

$i_{r,i}$: corresponding saturated air enthalpy evaluated at the inlet refrigerant temperature

$i_{r,o}$: corresponding saturated air enthalpy evaluated at the outlet refrigerant temperature



Basic Equations

Two-phase region

$$Q_{1c} = m_c \times i_{fg} \times (1 - x) = m_a \times (i_{a1} - i_{ao})$$

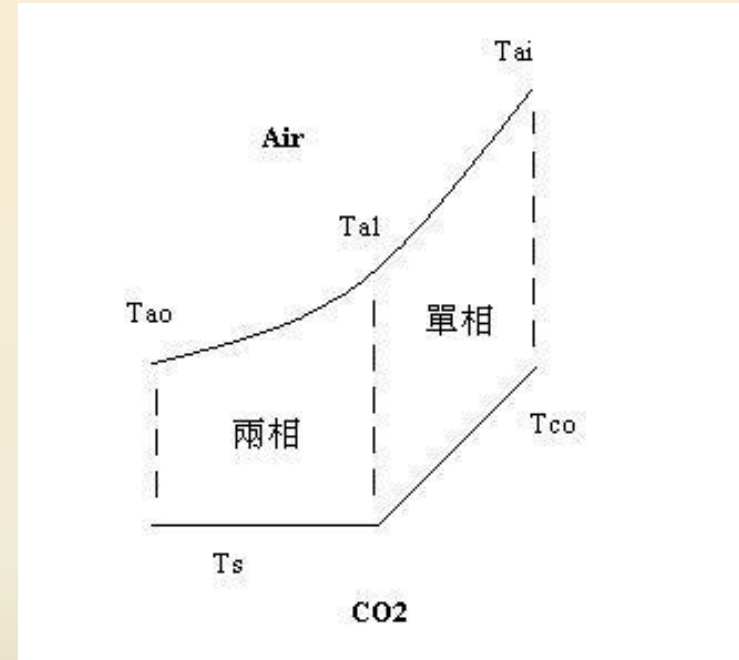
$$Q_{1c} = (UA)_1 \times (LMHD)_1$$

Single-phase region

$$Q_{2c} = m_c \times C_{p_c} \times (T_{co} - T_s) = m_a \times (i_{ai} - i_{a1})$$

$$Q_{2c} = (UA)_2 \times (LMHD)_2$$

Total heat exchange area $A_{total} = A_1 + A_2$





Evaporator

- The calculation of the air-side heat transfer coefficient
- Threlkeld (1970) Wet coil Analysis Method:

$$h_{o,w} = \frac{1}{\frac{c_{p,a}}{b'_{w,m} h_{c,o}}} = \left(\frac{c_{p,a}}{b'_{w,m} h_{c,o}} \right)^{-1}$$

$$b'_{w,m} = \frac{i - i_{s,w,m}}{T - T_{f,m}}$$

$$h_{c,o} = j_{wet} \times G_c \times C_{pa} \times \text{Pr}_a^{-2/3}$$



Evaporator

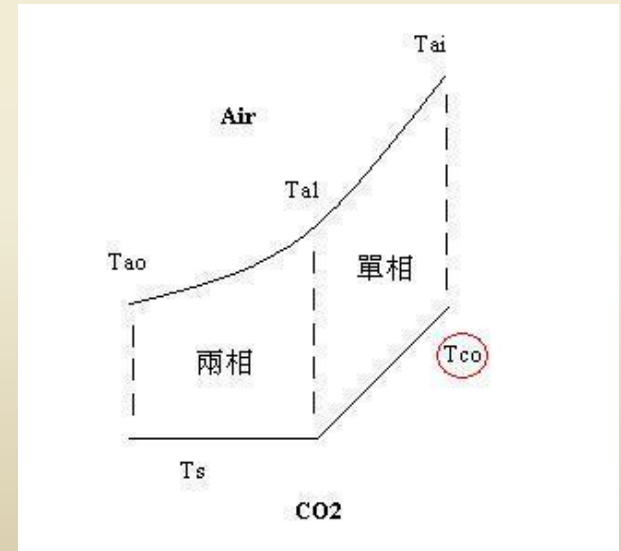
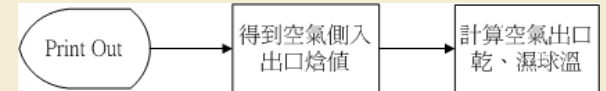
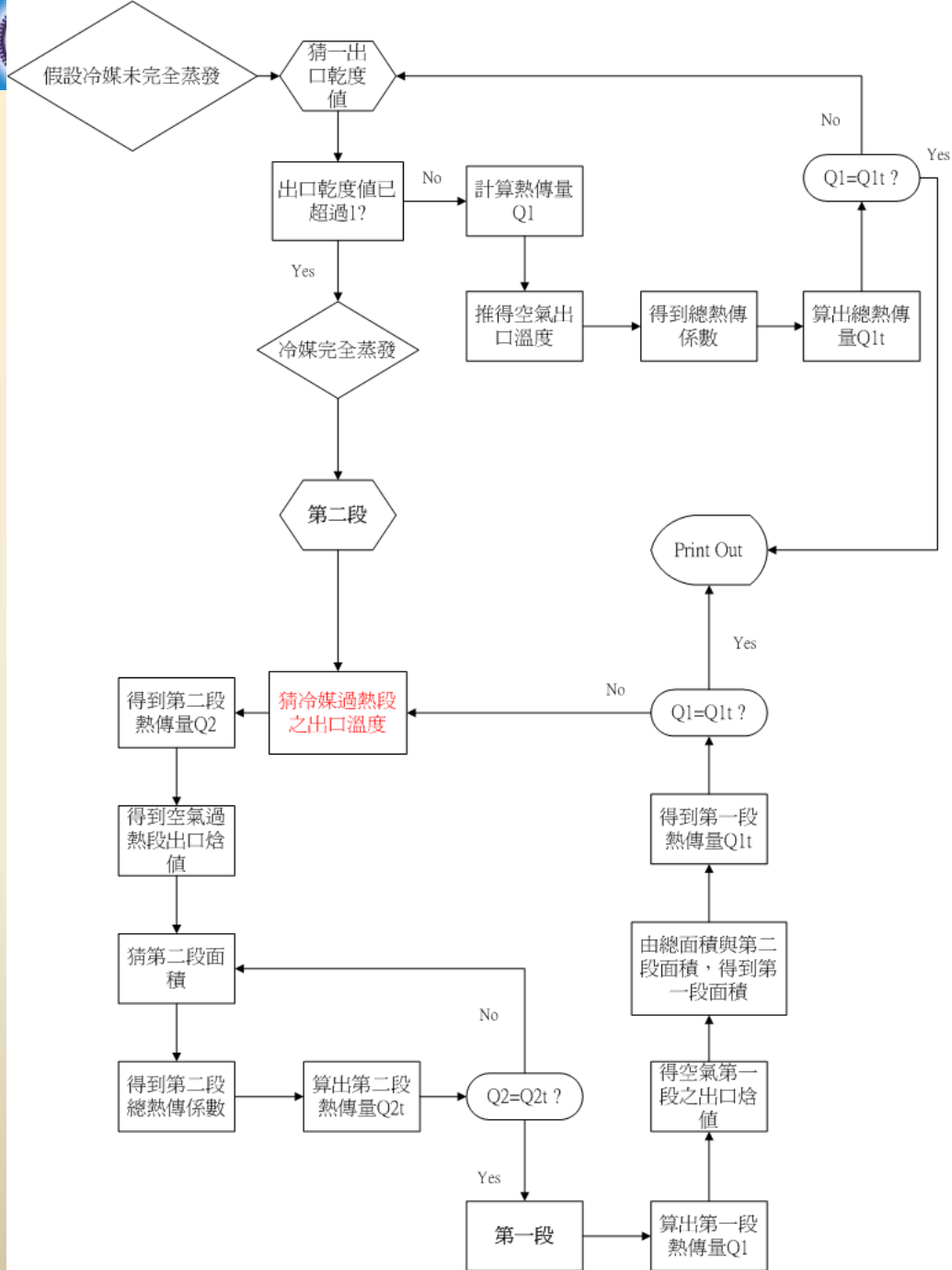
- Refrigerant side
- Hihara and Tanaka (2000) correlation :

$$\frac{\alpha}{\alpha_{lo}} = K_1 Bo + K_2 \left(\frac{1}{X_{tt}} \right)^{2/3}$$

$$\alpha_{lo} = 0.023 \text{Re}_l^{0.8} \text{Pr}_l^{0.4} \frac{\lambda_l}{d_i}$$

$$K_1 = 1.4 \times 10^4, \quad K_2 = 0.93$$

$$X_{tt} = \left(\frac{1-x}{x} \right)^{0.9} \left(\frac{\rho_g}{\rho_l} \right)^{0.5} \left(\frac{\mu_l}{\mu_g} \right)^{0.1}$$

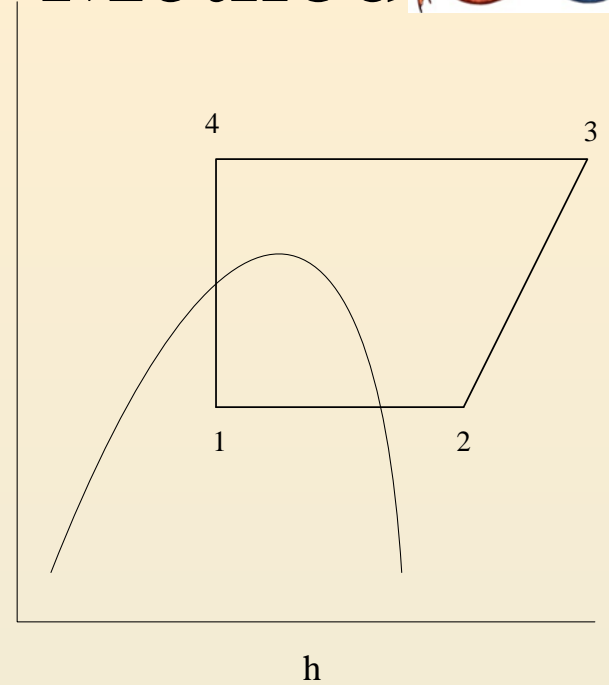




Cycle Analysis Method



The first step is assuming the compressor inlet pressure, inlet temperature, outlet pressure. With these three values, one can calculate the mass flow rate of refrigerant. After $2 \rightarrow 3$ adiabatic process, the discharge compressor temperature can be obtained. $3 \rightarrow 4$ One can obtain a outlet temperature and enthalpy using the gas cooler model. Thus the exit state after capillary tube can be obtained via the $4 \rightarrow 1$ using capillary model. Checking whether the outlet pressure is the same pressure as the original guess. If it is different, we must re-assume the compressor outlet pressure. $1 \rightarrow 2$ Entering the evaporator and calculate the heat transfer rate with air. If it does not satisfied the energy balance equation in evaporation region, we should re-assume the initial value from the first step.



Input parameters

Gas cooler dimensions (d_i, d_o, D_i, L)

Evaporator dimensions ($d_i, d_o, D_i, W, H...$)

Expansion device dimensions (d_c, L_c)

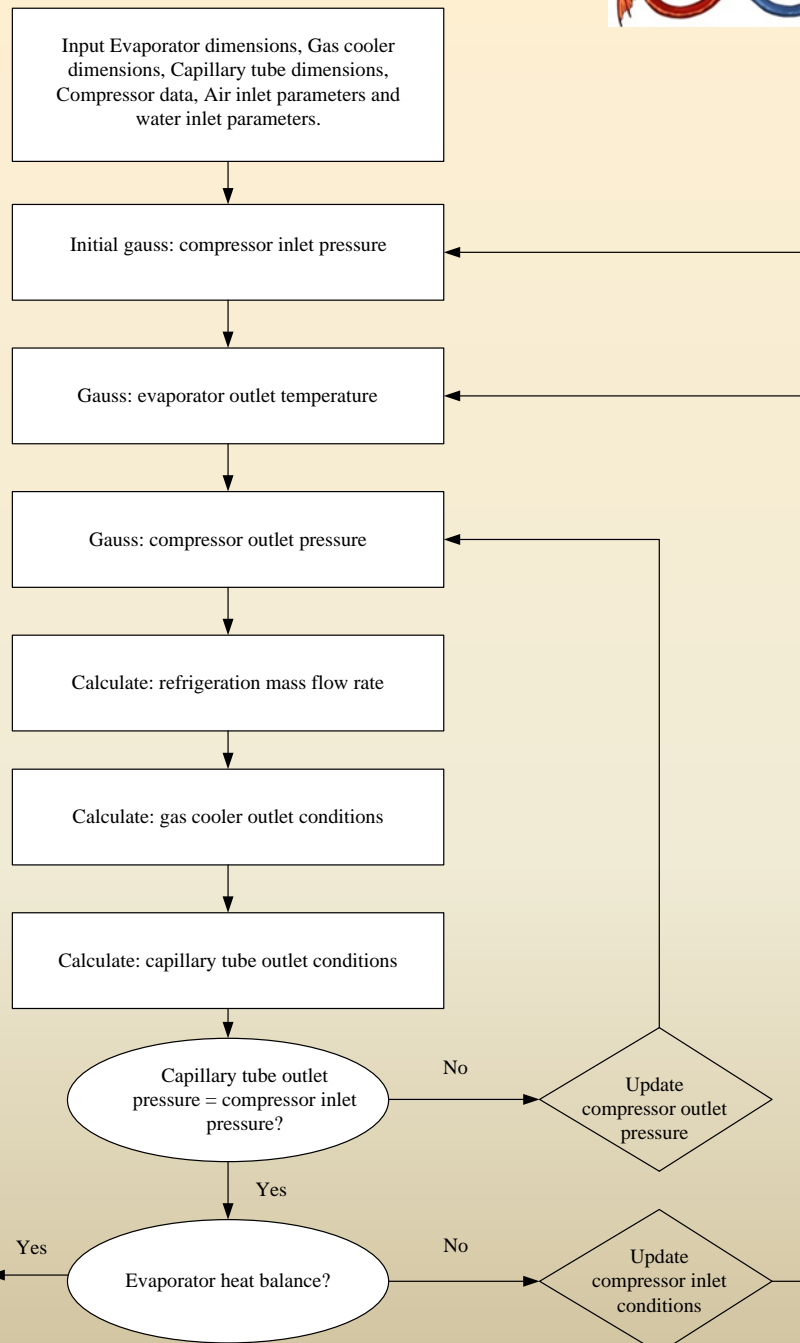
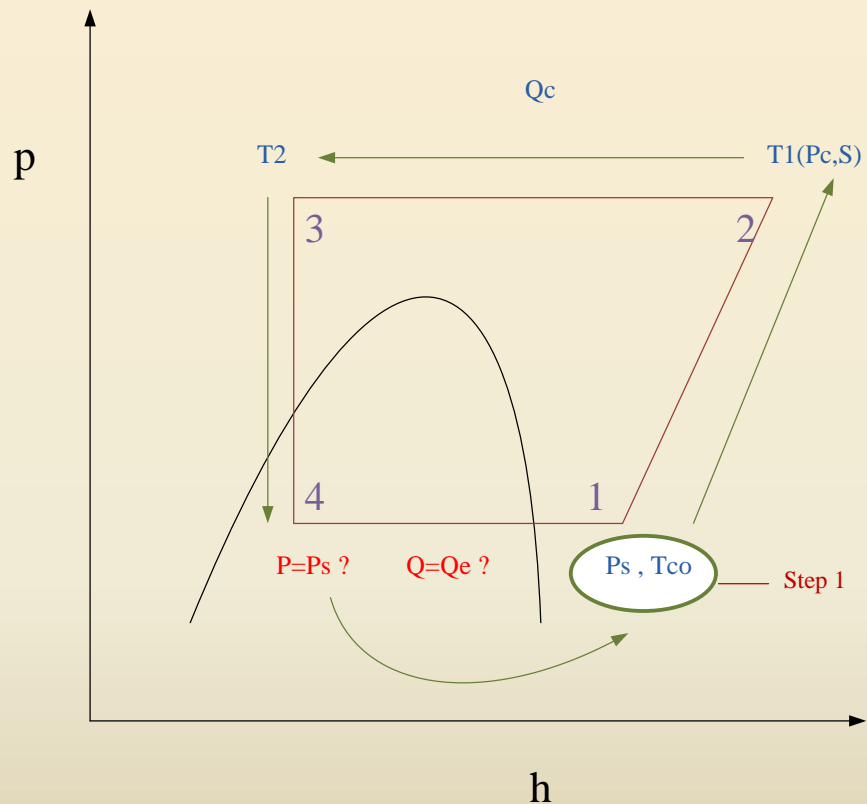
Compressor data: V_s, N

Water inlet conditions (m_w, T_{wi})

Air inlet conditions (m_a, T_{db}, T_{wb})

The calculation process make use the following assumptions:

1. Heat loss is negligible
2. Steady state
3. No pressure drop in gas cooler & evaporator
4. Ignore the change of kinetic and potential energy in capillary tube

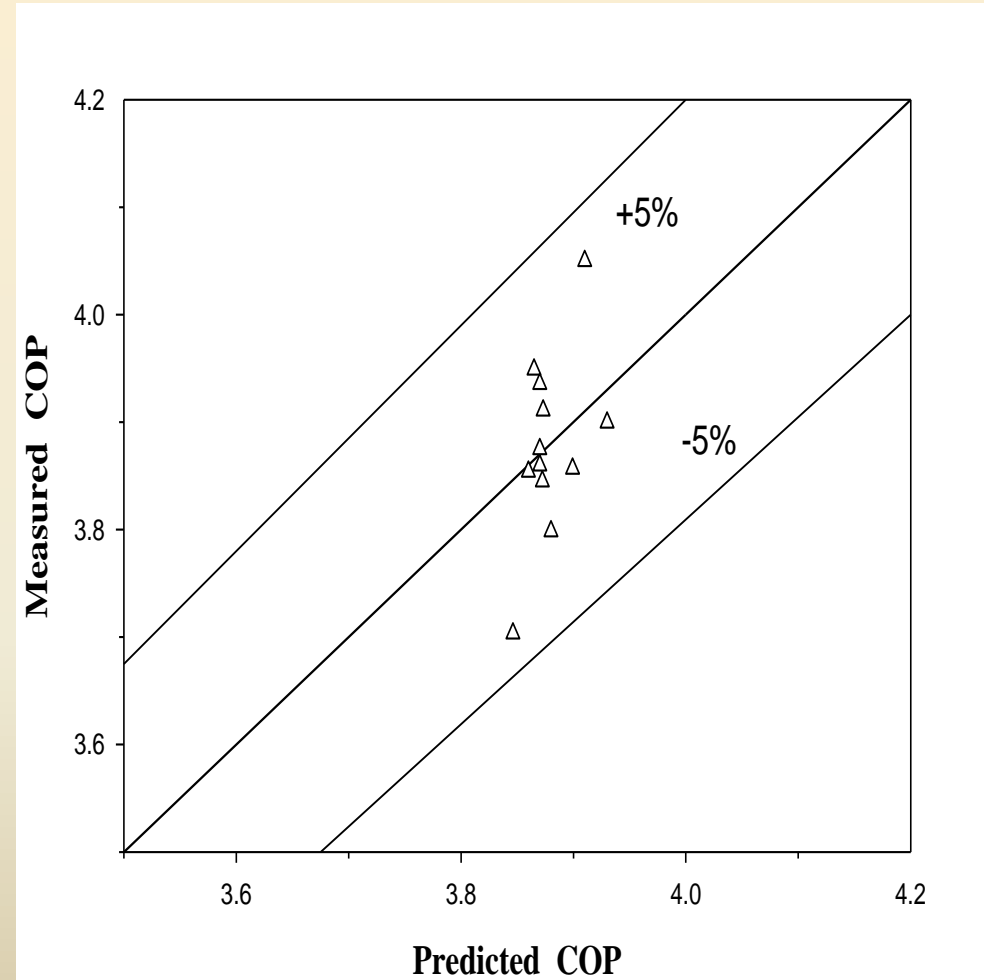
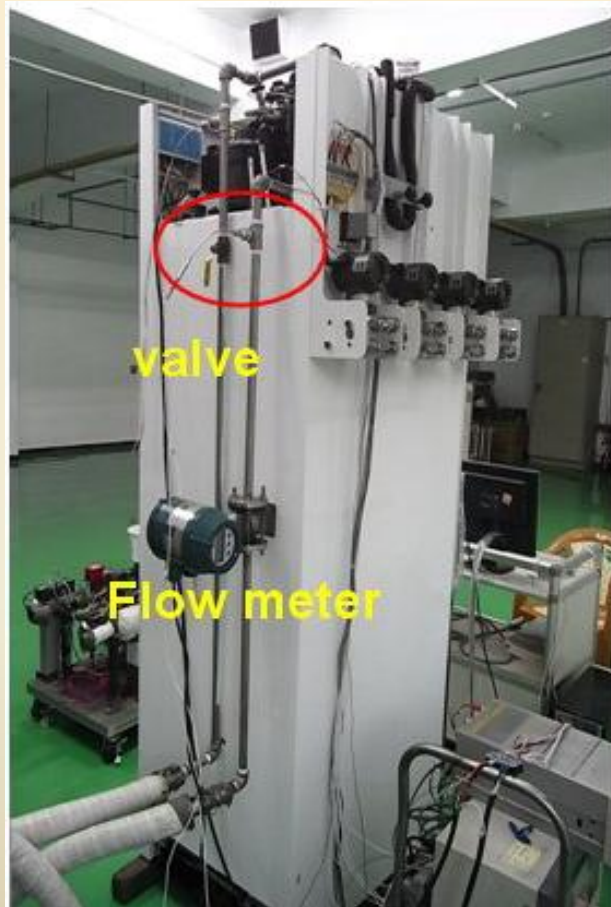




Simulation results

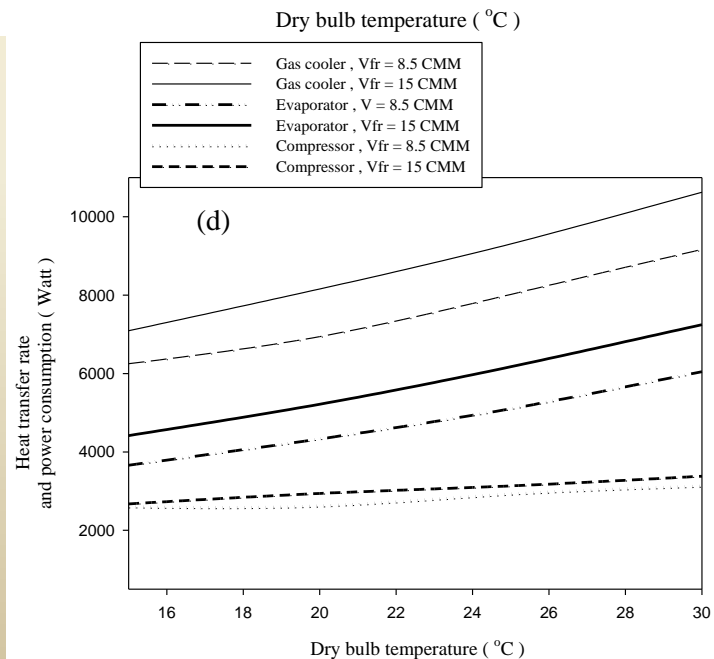
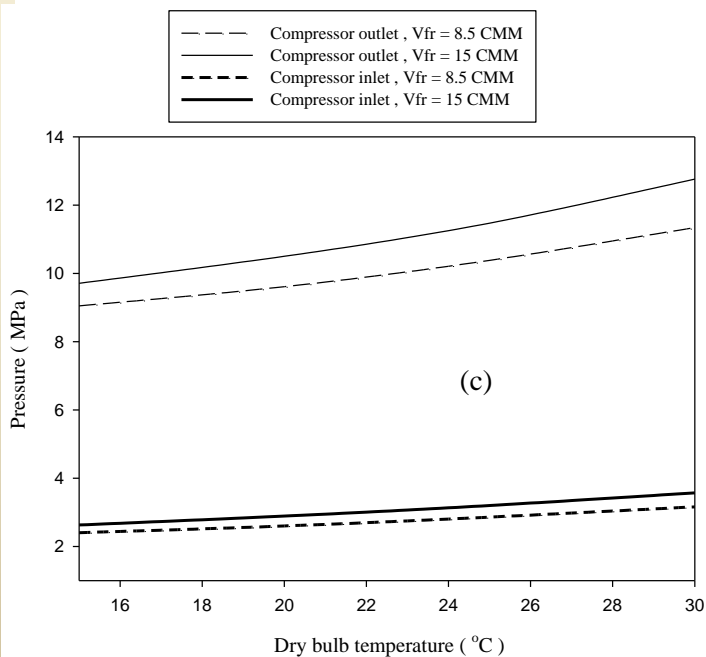
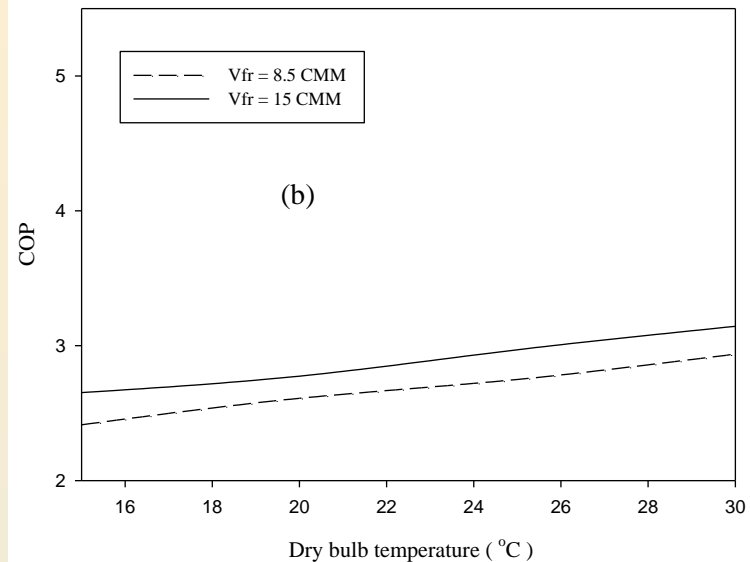
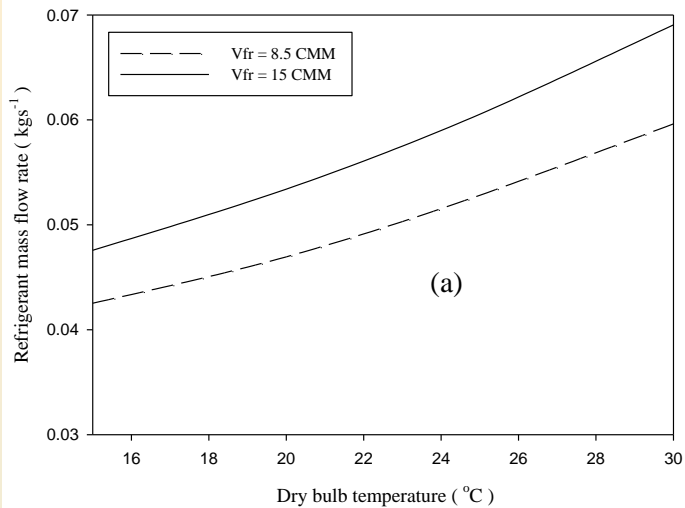


Predictions vs. Measurements



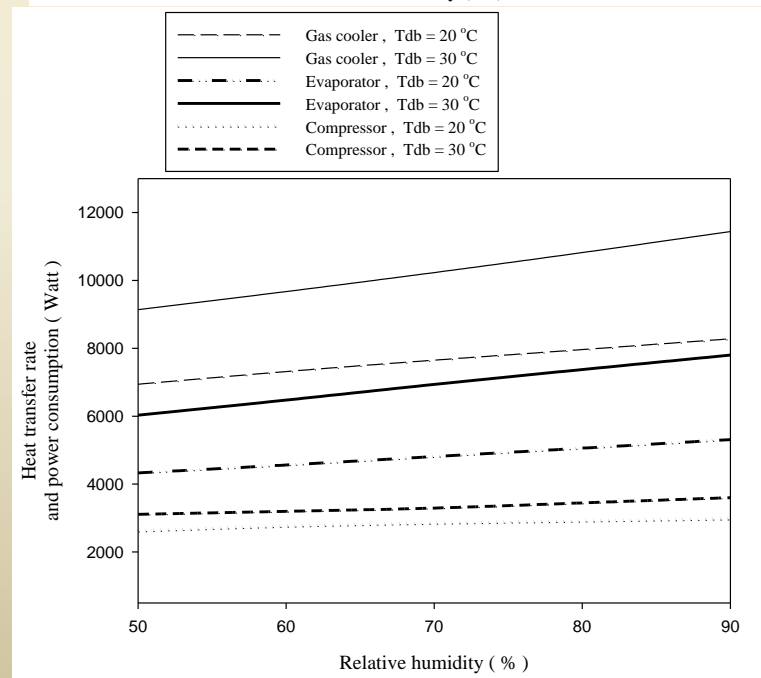
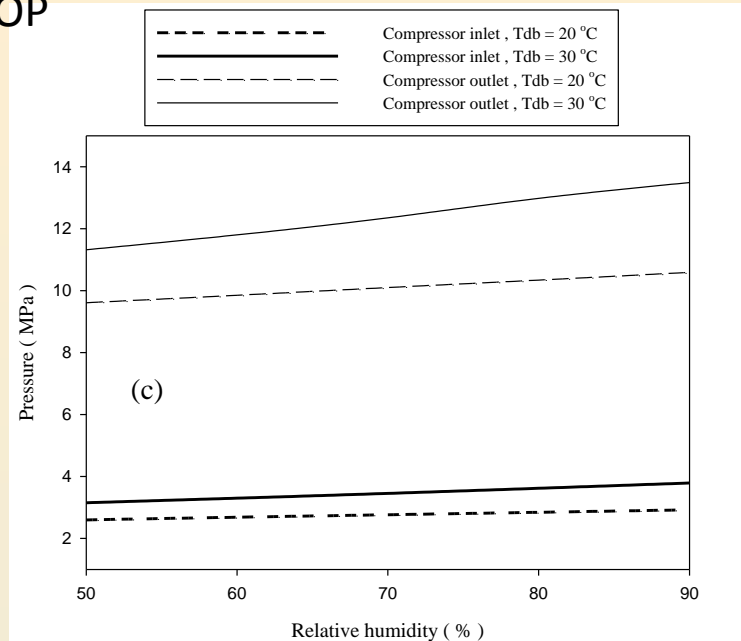
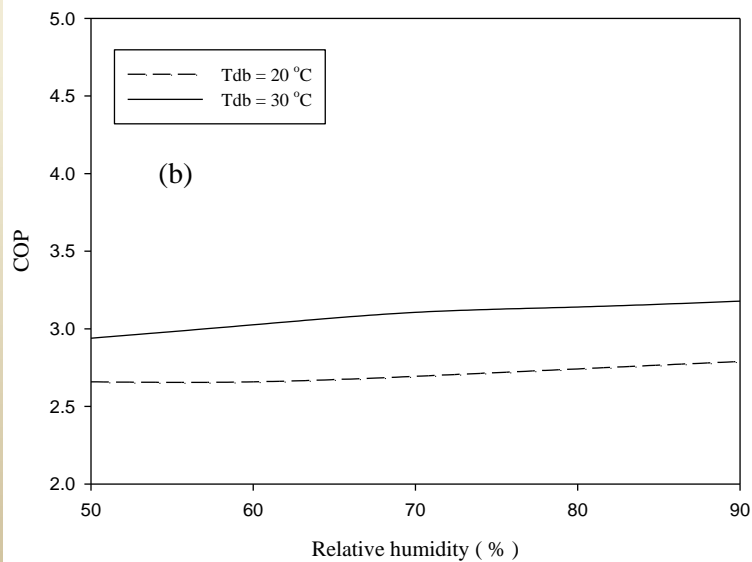
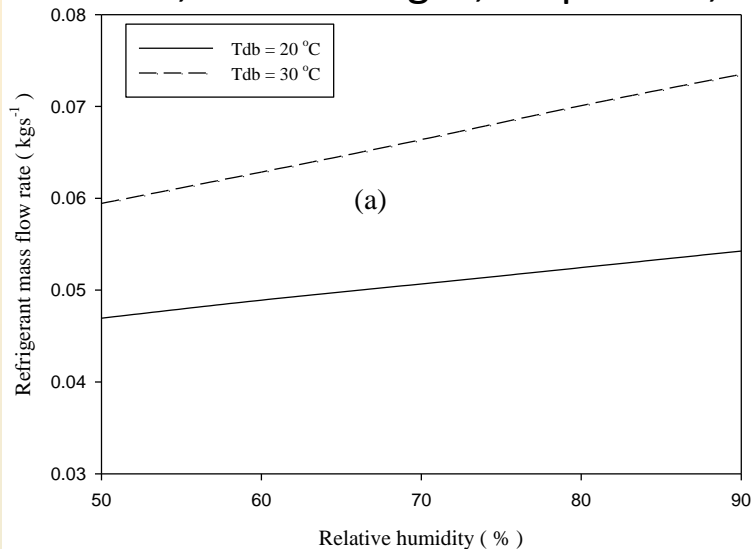


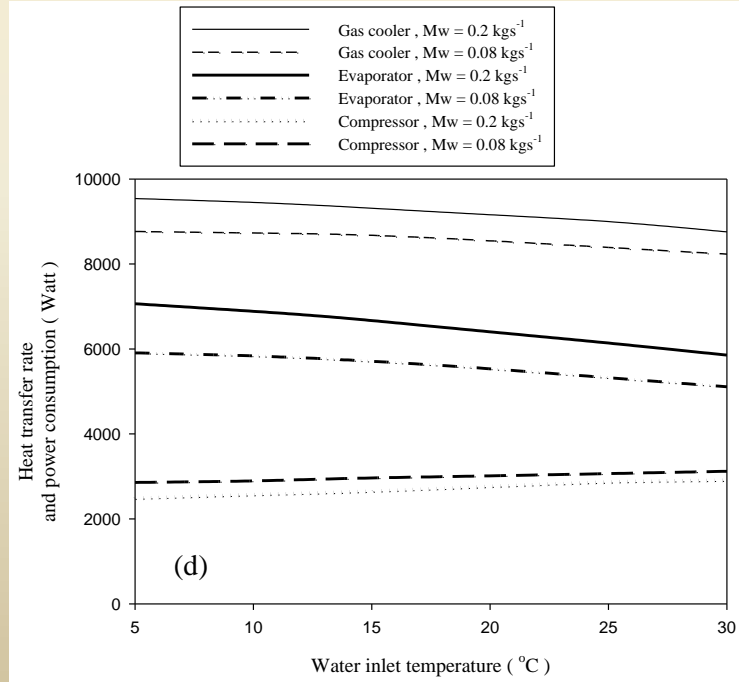
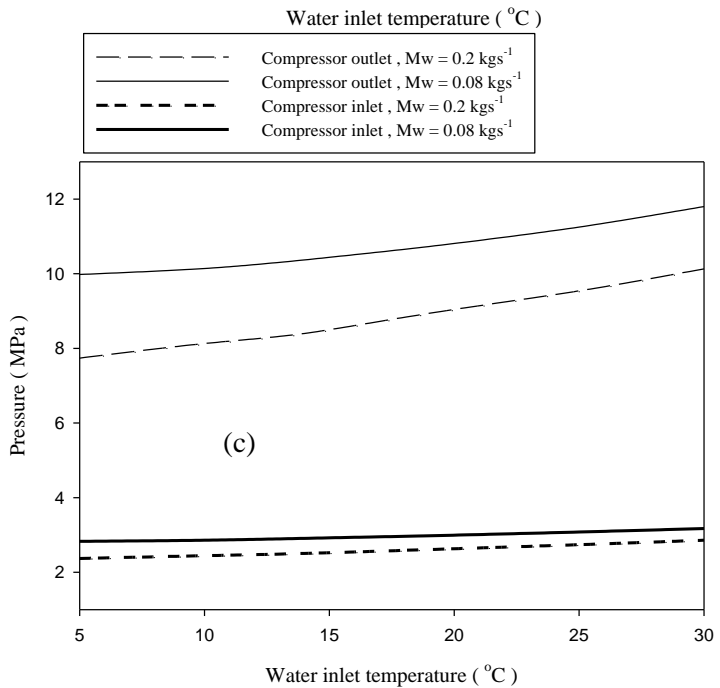
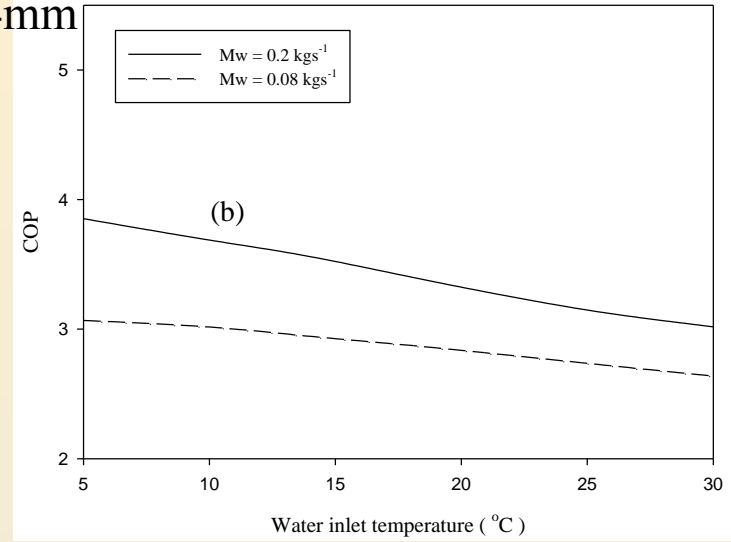
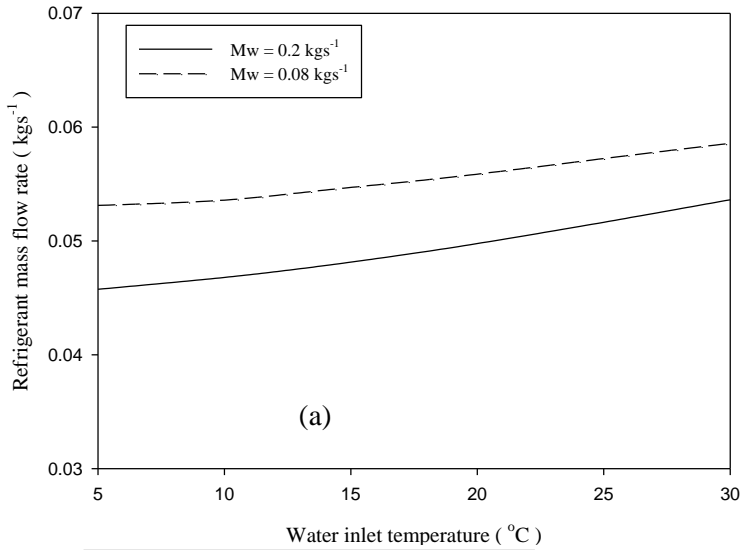
Effect of the dry bulb air temperature on (a) refrigerant mass flow rate (b) COP (c) pressure (d) heat transfer rate. With RH=50%, $T_{wi}=20^{\circ}\text{C}$, $M_w=0.08\text{kgs}^{-1}$, $L_{cap}=2.0\text{m}$, $d_{,cap}=1.4\text{mm}$





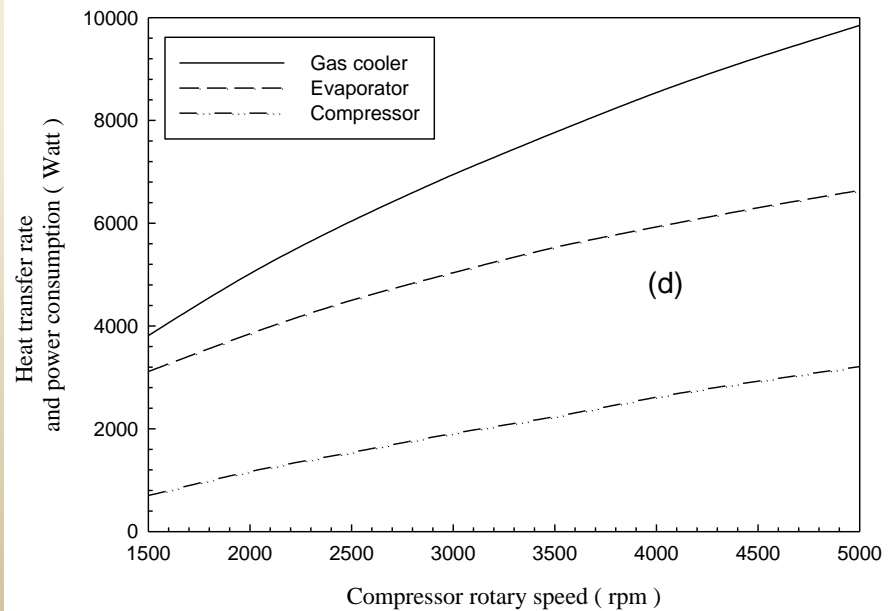
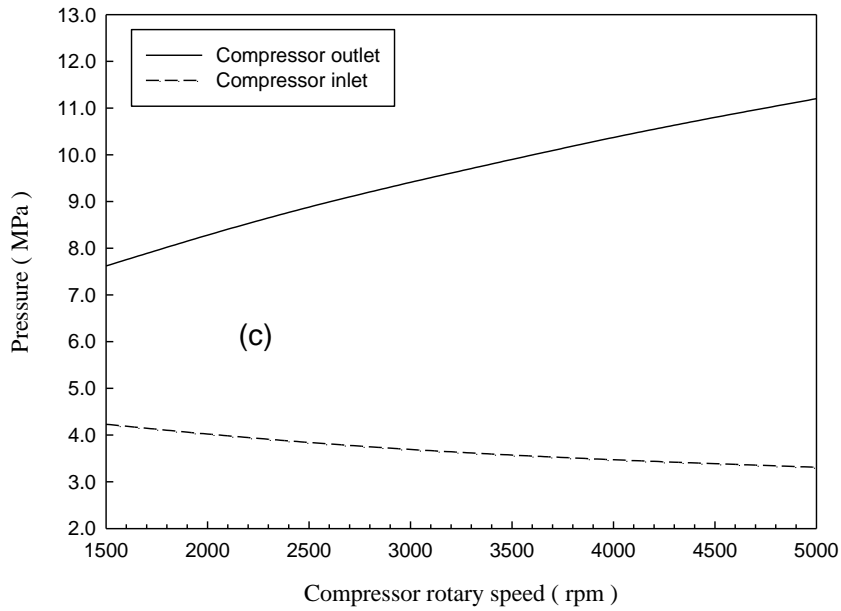
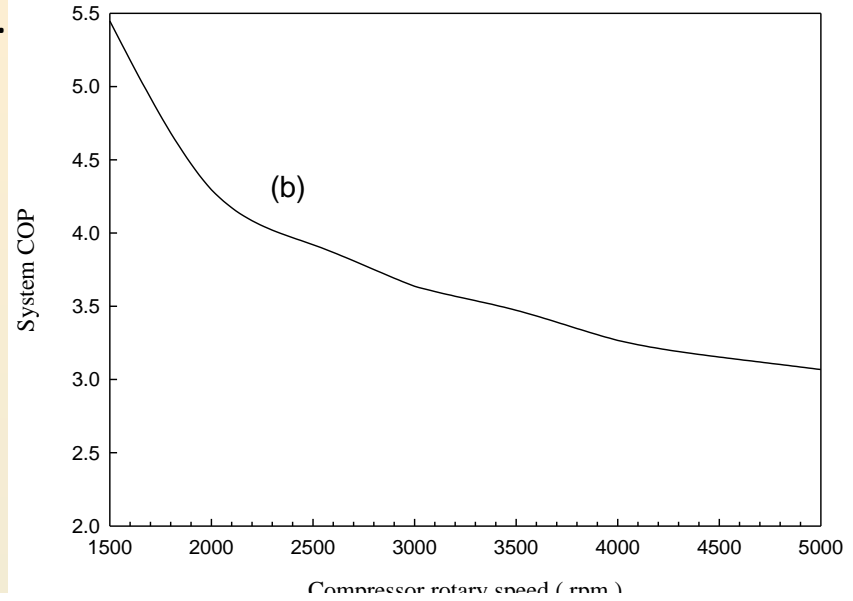
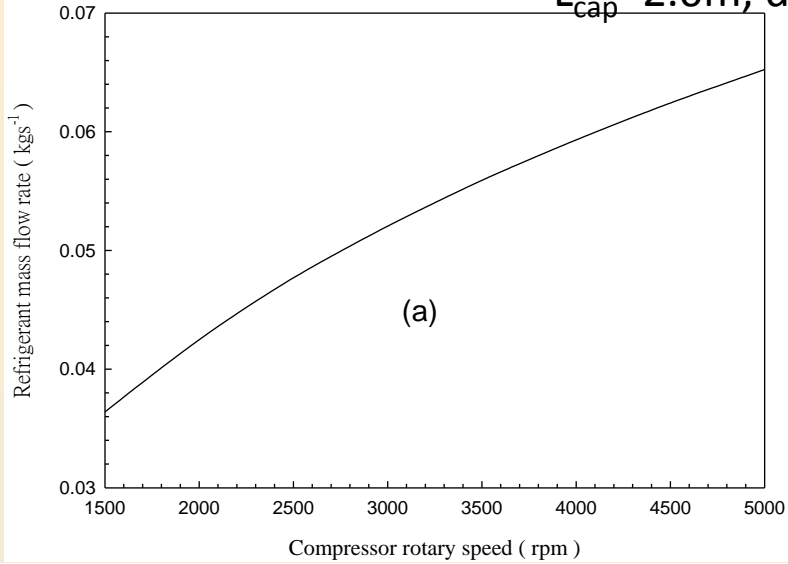
Effect of the RH on the (a) refrigerant mass flow rate (b) COP (c) pressure (d) heat transfer rate. With $V = 8.5 \text{ CMM}$, $T_{wi} = 20^\circ\text{C}$, $M_w = 0.08 \text{ kgs}^{-1}$, $L_{cap} = 2.0 \text{ m}$, $d_{cap} = 1.4 \text{ mm}$,







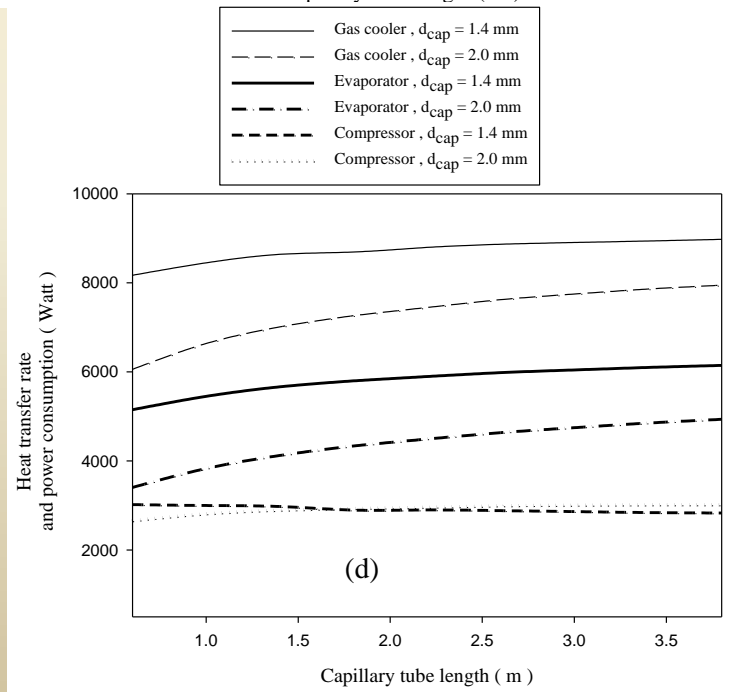
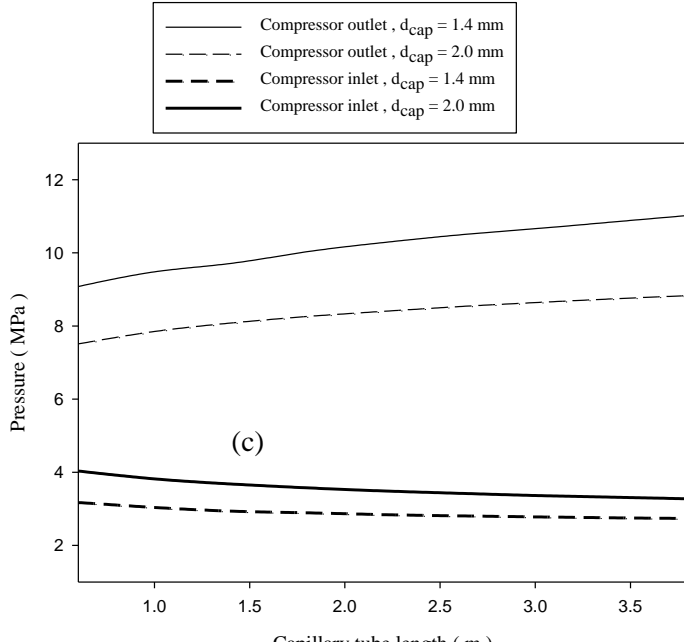
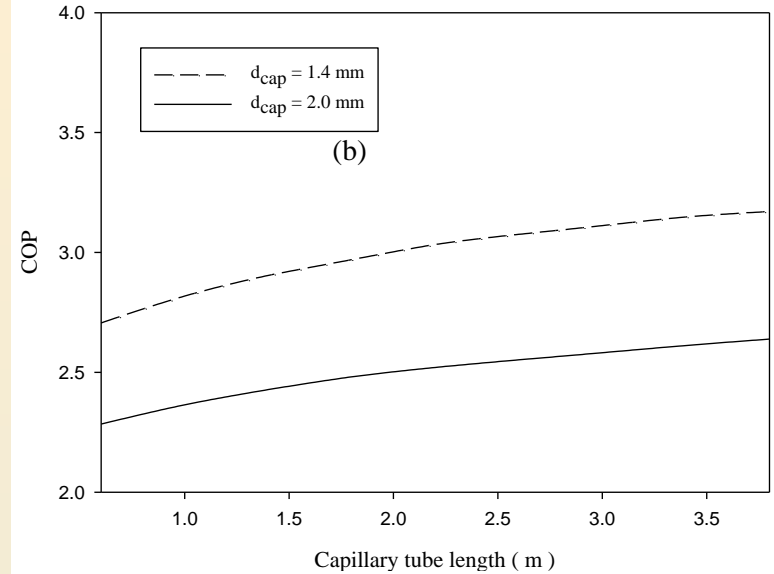
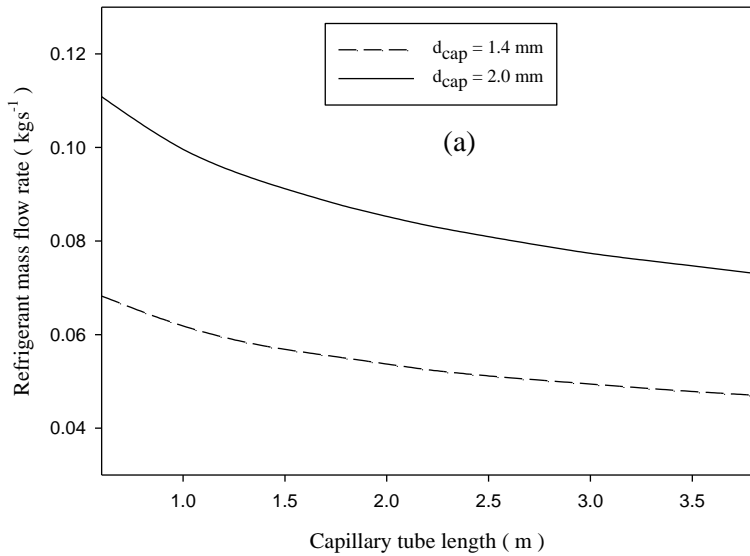
Effect of compressor speed on (a) refrigerant mass flow rate (b) COP (c) pressure (d) heat transfer rate. With $T_{db} = 27^\circ\text{C}$, $T_{wb} = 20^\circ\text{C}$, $V = 8.5\text{CMM}$, $T_{wi} = 10^\circ\text{C}$, $M_w = 0.2\text{kgs}^{-1}$, $L_{cap} = 2.0\text{m}$, $d_{cap} = 1.4\text{mm}$.





Effect of the capillary tube length on (a) refrigerant mass flow rate (b) COP (c) pressure (d) heat transfer rate.

With $T_{db}=27^{\circ}\text{C}$, $T_{wb}=20^{\circ}\text{C}$, $V=8.5\text{CMM}$, $T_{wi}=10^{\circ}\text{C}$, $M_w=0.2\text{kgs}^{-1}$





Conclusions

- This study develops a simulation program of CO₂ refrigerant cycle system. The simulation program is capable of handling the variation of indoor conditions, gas cooler, compressor speed, and geometry of capillary tube without any prescribed conditions (i.e. fixed evaporation temperature, fixed condensing temperature, and the like).
- In gas cooler, the CO₂ may present a local minimum and a local maximum along the length of the heat exchange, provided CO₂ passes through the pseudo-critical temperature, and this phenomenon becomes more and more pronounced when the pressure is close to the critical pressure
- The RH of the indoor air is an important factor affecting the overall system performance.
- Increasing the water side inlet temperature or increasing the compressor speed will decrease COP of the system.
- An increase of the length of capillary tube also increases the COP moderately.



Thanks for your attention



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